1. Base your answer to the following question on Given the nuclear equation:

4H + X - + 6dLi + 4He

The particle represented by X is

- A) % Li B) % Be C) 10 Be D) 10 C
- 2. Which isotope will spontaneously decay and emit particles with a charge of +2?
  - A) 53Fe B) 137Cs C) 198Au D) 220Fr
- 3. Which of these particles has the greatest mass?
  - A) alpha

B) beta

C) neutron

- D) positron
- 4. Base your answer to the following question on Given the nuclear equation:

1910No - X + 190F

What particle is represented by X?

A) alpha

B) beta

C) neutron.

- D) positron.
- 5. Alpha particles and beta particles differ in
  - A) mass, only
  - B) charge, only
  - C) both mass and charge
  - D) neither mass nor charge
- 6. Which statement best describes gamma radiation?
  - A) It has a mass of 1 and a charge of 1.
  - B) It has a mass of 0 and a charge of -1.
  - C) It has a mass of 0 and a charge of 0.
  - D) It has a mass of 4 and a charge of +2.
- Which type of radiation is most similar to high-energy x-rays?
  - A) alpha

B) beta

C) neutron

- D) gamma
- What is the total number of years that must pass before only 25.00 grams of an original 100.0-gram sample of C-14 remains unchanged?
  - A) 2865 y

B) 5730 y

C) 11 460 y.

- D) 17 190 v
- 9. Which isotope is most commonly used in the radioactive dating of the remains of organic materials?
  - AN PAC
- B) 16N
- CD 33P
- D) 37K

- 10Based on Reference Table N, what fraction of a radioactive 90 Sr sample would remain unchanged after 56.2 years?

- A)  $\frac{1}{6}$  B)  $\frac{1}{4}$  C)  $\frac{1}{8}$  D)  $\frac{1}{16}$
- 11. What is the half-life and decay mode of Rn-222?
  - A) 1.91 days and alpha decay
  - B) 1.91 days and beta decay
  - C) 3.82 days and alpha decay
  - D) 3.82 days and beta decay
- 12. After 32 days, 5 milligrams of an 80-milligram sample of a radioactive isotope remains unchanged. What is the half-life of this element?
  - A) 8 days

B) 2 days

C) 16 days.

- D) 4 days
- 13. Which reaction converts an atom of one element to an atom of another element?
  - A) combustion
- B) polymerization
- C) saponification
- D) transmutation
- Radioactive cobalt-60 is used in radiation therapy treatment. Cobalt-60 undergoes beta decay. This type of nuclear reaction is called
  - A) natural transmutation
  - B) artificial transmutation
  - C) nuclear fusion.
  - D) nuclear fission
- 15. In which reaction is mass converted to energy by the process of fission?
  - A)  $^{14}{}_{2}N + ^{1}{}_{0}n \rightarrow {}^{14}{}_{6}C + ^{1}{}_{1}H$
  - B)  $^{235}_{92}U + ^{1}_{9}n \rightarrow ^{87}_{18}Br + ^{146}_{52}La + 3^{1}_{9}n$
  - C) <sup>226</sup><sub>88</sub>Rn → <sup>222</sup><sub>86</sub>Rn + <sup>4</sup><sub>2</sub>He
  - D)  $^{2}(H + ^{2})H \rightarrow ^{4}2He$
- Which statement best describes what happens in a fission reaction?
  - A) Heavy nuclei split into lighter nuclei.
  - B) Light nuclei form into heavier nuclei.
  - C) Energy is released and less stable elements are formed.
  - Energy is absorbed and more stable elements are formed.

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**Bowker Editorial Staff** 

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