Name: _____ Date: _____

Electrons in Atoms Study Worksheet

Fill in the blanks in the foll	owing paragraph with the o	ppropriate terms.
In Bohr's model of the ato	m, electrons inhabit certain _	levels,
with the levels closest to	the nucleus having	energy than those
farther from the nucleus. I	n the state of	the atom, the electrons
are in the lowest	level possible. When an	atom absorbs energy,
it is said to ascend to an _	state, where it i	s unstable. The atom
will soon the s	ame amount of energy absor	bed which may be seen
in the form of visible light.	. In the study of	, this visible light
is seen as the	spectrum of an e	lement. This spectrum is
to each ele	ment, hence it is also called	an element's
"fingerprints". The moderr	n view of light is that it has a	nature.In
other words, light may bel	have as a stream of particles	calledor
, or light i	may behave as a	-
In the wave view of light, t	he wave equation is often us	ed to determine a
wave's frequency or wavel	ength. The	is the distance
between corresponding po	oints on adjacent waves whil	e the
is the number of waves pa	ssing a given point in a given	time. The wave
equation is:		

Chapter 5 Electrons In Atoms Worksheet

Chandan Sengupta

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