
Civics EOC Exam Study Guide Questions And Correct Answers 2023 Graded A+(verified)

- 1. What ideas is John Locke known for?:** natural rights, natural law, and social contract
- 2. Social Contract:** people give away some of their power to the government in exchange for protection
- 3. How did Montesquieu influence the Founding Fathers?:** by giving them the ideas of Separation of Powers and Checks and Balances
- 4. 3 concepts found in the Magna Carta that influenced the Founding Fathers :** trial by jury, limited government, rule of law
- 5. 3 rights that are found in the English Bill of rights that are found in the US Bill of Rights:** freedom of speech, no cruel and unusual punishment, no taxes without consent
- 6. 3 concepts in the Mayflower Compact that influenced the Founding Fathers-**
: self-government, rule of law, and social contract

Civics Study Guide

A Loxley



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web by utilizing the half equivalence point on the titration curve the pka of the acid analyte was determined to be 3.655 furthermore using the stoichiometric ratio between the naoh and the acid analyte 1:1 we could experimentally determine the number of moles of the acid in the solution to be 0.00731 moles making the experimental molar mass

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web goals understand that parts of an acid base titration be able to determine the K_a or K_b from pH data associated with the titration of a weak acid or base be able to determine the molar mass of a solid monoprotic acid from titration data be able to calculate K_{a1} and K_{a2} for a polyprotic acid by the end of this lab students should be able to

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web titration experiment in the neutralization of hydrochloric acid by sodium hydroxide the mole ratio of acid to base is 1:1

$\text{hcl aq} + \text{naoh aq} \rightarrow \text{nacl aq} + \text{h}_2\text{O l}$ one mole of hcl would be fully neutralized by one mole of naoh

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web conclusion the identification of an unknown solid acid the purpose of this experiment was to determine the standardization of sodium hydroxide naoh and oxalic acid $\text{H}_2\text{C}_2\text{O}_4 \cdot 2\text{H}_2\text{O}$ and the molar mass of an unknown solid acid using the molar mass of the standardized sodium hydroxide solution

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web conclusion this lab evaluated an acid base reaction with the volumetric analysis of a standard solution being naoh reaction with a 2nd substance $\text{H}_2\text{C}_2\text{O}_4 + 2\text{NaOH} \rightarrow \text{Na}_2\text{C}_2\text{O}_4 + 2\text{H}_2\text{O}$ to differentiate an acid from a base in a solution a pH scale is used the pH scale ranges from 0 to 14 a pH of 7 is neutral in an acid base reaction

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ascertain the concentration of the NaOH

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web this point of titration is referred to as the equivalence point considering the 1:1 stoichiometry of this acid base reaction

$\text{NaOH (aq)} + \text{C}_6\text{H}_4\text{COOH (aq)} \rightarrow \text{C}_6\text{H}_4\text{COONa (aq)} + \text{H}_2\text{O (l)}$ the point of equivalence is the point of titration when the number of moles of NaOH added is equal to the number of moles of KHP in the