CLASSIFYING CHEMICAL REACTIONS

Part II

Classify the following reactions as synthesis, decomposition, single replacement or double replacement.

ANSWERS

1.	$2KClO_3 \rightarrow 2KCl + 3O_2$	Decomposition
2.	$HCl + NaOH \rightarrow NaCl + H_2O$	Double Replacement
3.	$Mg + 2HCl \rightarrow MgCl_2 + H_2$	Single Replacement
4.	$2H_2 + O_2 \rightarrow 2H_2O$	Synthesis
5.	$2Al + 3NiBr_2 \rightarrow 2AlBr_3 + 3Ni$	Single Replacement
6.	$4Al + 3O_2 \rightarrow 2Al_2O_3$	Synthesis
7.	$2NaCl \rightarrow 2Na + Cl_2$	Decomposition
8.	$CaCl_2 + F_2 \rightarrow CaF_2 + Cl_2$	Single Replacement
9.	$AgNO_3 + KCl \rightarrow AgCl + KNO_3$	Double Replacement
10.	$N_2 + 3H_2 \rightarrow 2NH_3$	Synthesis
11.	$2H_2O_2 \rightarrow 2H_2O + O_2$	Decomposition
12.	$(NH_4)_2SO_4 + Ba(NO_3)_2 \rightarrow BaSO_4 + 2NH_4NO_3$	Double Replacement
13.	$MgI_2 + Br_2 \rightarrow MgBr_2 + I_2$	Single Replacement
14.	$SO_3 + H_2O \rightarrow H_2SO_4$	Synthesis

Classifying Chemical Reactions Study Guide Answers

Karen Timberlake

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