

Classifying Chemical Reactions

Analyzing and Predicting Products

Name _____

Background:

In a chemical equation, the reactants are written on the left side of the equation, the products on the right side of the equation. An arrow represents the direction of the reaction and is read as "yields" or "produces."

Chemical reactions can be classified by considering the number of reactants and products in the reaction, the physical or chemical nature of the reactants and products, and the how the atoms are rearranged during the reaction. The table below summarizes some of the most common reaction types. Notice that none of the reactions are balanced; you balance these reactions as part of the pre-lab activities.

Types of Chemical Reactions	
Synthesis	Two reactants combine to form a single product. The reactants may be elements or compounds. Also called a synthesis reaction. $\text{___ Na (s) + ___ Cl}_2 \text{ (g) } \rightarrow \text{___ NaCl (s)}$
Decomposition	One complex reactant breaks down to give two or more simpler products. $\text{___ H}_2\text{O}_2 \text{ (l) } \rightarrow \text{___ H}_2\text{O (l) + ___ O}_2 \text{ (g)}$
Single Replacement	An element reacts with a compound and replaces one of the elements in the compound. Elements with positive oxidation numbers must replace atoms with a positive oxidation. The same is true for elements with negative oxidation numbers.
Double Replacement	Two ionic compounds exchange ions to form new compounds. $\text{___ CuCl}_2 \text{ (aq) + ___ Na}_3\text{PO}_4 \text{ (aq) } \rightarrow \text{___ Cu}_3\text{(PO}_4\text{)}_2 \text{ (s) + ___ NaCl (aq)}$
Combustion	A compound burns in the presence of oxygen, producing heat and light. The combustion of organic compounds (compounds made of carbon) produces carbon dioxide and water. $\text{___ C}_8\text{H}_{18} \text{ (l) + ___ O}_2 \text{ (g) } \rightarrow \text{___ CO}_2 \text{ (g) + ___ H}_2\text{O (l)}$

Pre-Lab Questions and Activities:

1. Balance the equations in the previous table.

Classifying Chemical Reactions Analyzing And Predicting

**Jorge M. C. Marques, Emilio Martinez-
Nunez, William L. Hase**



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