

(c) The reaction between NO and O_3 is as shown.



Investigations of the kinetics of the reaction were carried out by measuring the initial rates of decrease in pressure of known mixtures of the gases. The following results were obtained.

	expt 1	expt 2	expt 3
Initial pressure of NO/atm	1.807	1.807	0.904
Initial pressure of O_3 /atm	4.673	3.117	4.673
Initial rate of decrease of pressure/atm min ⁻¹	0.500	0.333	0.165

- Determine the order of reaction with respect to NO and O_3 .
- Write the rate expression for the reaction.
- Use experiment 1 to calculate the rate constant and state its units.
- State, by means of equations, how the reaction takes part in the catalysis of acid rain formation.

(8)
(Total: 12)

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Yan Bai



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