

## Practice Problems

(Answers following problems)

### SECTION 18.1 PROPERTIES OF SOLUTIONS

1. The following table summarizes some data on the vapor pressures of glycerol at 5 mm Hg. Assume that the data are correct.

(Problem 2)

2. The vapor pressure of the liquid phase of glycerol is 1.00 mm Hg at 25°C. The vapor pressure of the solid phase of glycerol is 0.10 mm Hg at 25°C. Calculate  $\Delta H_{\text{fusion}}$  for glycerol.

3. The following table shows some data regarding the boiling points of some liquids. Assume that the data are correct. Calculate the  $\Delta H_{\text{vaporization}}$  for each liquid.

(Problem 3)

### SECTION 18.2 CONCENTRATIONS OF SOLUTIONS

1. Calculate the mass of solute in each of the following solutions.

- a. 2.00 mL of a 0.100 M solution of NaCl in water

(Problem 1)

- b. 5.00 g of a 10.0% solution of NaCl in water

(Problem 2)

2. Calculate the volume of pure solvent needed to prepare each of the following solutions.

- a. 100.0 mL of a 0.100 M solution of NaCl in water

(Problem 3)

- b. 100.0 mL of a 10.0% solution of NaCl in water

(Problem 4)

3. Calculate the molarity of a solution of NaCl in water if 5.00 g of NaCl is dissolved in 100.0 mL of water.

(Problem 5)

# Chemistry Ch 16 Solutions Practice Problems Answers

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