

Name: _____

Date: _____

Significant Figures

Determine the number of significant figures in each of these numbers.

	Number	Significant Figures		Number	Significant Figures
1.	357	_____	2.	10000	_____
3.	51015	_____	4.	6.060×10^{-2}	_____
5.	0.0007	_____	6.	4.556×10^{-9}	_____
7.	5050	_____	8.	5050.0	_____
9.	6.8×10^3	_____	10.	0.002110	_____
11.	33.303	_____	12.	170	_____
13.	15.0×10^{-5}	_____	14.	0.7007	_____
15.	0.70070	_____	16.	4206	_____
17.	0.02	_____	18.	10.01	_____
19.	0	_____	20.	0.0	_____

Chemistry Chapter 1 Significant Figures Worksheet

M Walker



Chemistry Chapter 1 Significant Figures Worksheet:

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Chemistry James N. Spencer, George M. Bodner, Lyman H. Rickard, 2010-12-28 **CHEMISTRY Students Solutions Manual to Accompany Physical Chemistry: Quanta, Matter, and Change 2e** Charles Trapp, Marshall Cady, Carmen Giunta, 2014 The Students Solutions Manual to Accompany Physical Chemistry Quanta Matter and Change 2e provides full worked solutions to the exercises and the odd numbered discussion questions and problems presented in the parent book The manual is intended for students and instructors alike and provides helpful comments and friendly advice to aid understanding

Fundamentals of Analytical Chemistry Douglas A. Skoog, 2004 This text is known for its readability combined with a systematic rigorous approach Extensive coverage of the principles and practices of quantitative chemistry ensures suitability for chemistry majors

Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction

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Pesce, Steven C. Kazmierczak, 2003 This comprehensive up to date readable text acts as a complete clinical chemistry course and professional reference providing detailed specific information on the principles of clinical chemistry in laboratory diagnosis as well as the pathophysiologic changes that occur in disease and affect testing outcomes Explanations of Laboratory Techniques Part 1 lead the reader through various necessary laboratory techniques and practices Chapters on Pathophysiology Part 2 provide descriptions of how specific diseases affect the human body A companion CD ROM packaged with the book features Methods of Analysis a comprehensive Urinalysis Manual and an interactive Study Guide Workbook to reinforce concepts The book s clear writing and comprehensive coverage make it an ideal resource for both students and practitioners Instructor resources are available to qualified adopters contact your sales representative for more information

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