

CK-12 Chemistry Concepts - Intermediate Answer Key

Chapter 13: States of Matter

13.1 Kinetic-Molecular Theory

Practice

Questions

Use the link below to answer the following questions:

http://www.chem.ufl.edu/~itl/2045/lectures/lec_d.html

1. Who developed the kinetic-molecular theory?
2. How many molecules are involved in most collisions?
3. What assumption is made about the volume of a mole of an ideal gas?

Answers

1. Rudolf Clausius.
2. Two.
3. The volume is essentially zero.

Review

Questions

1. What kind of motion are gases in?
2. What kind of collisions occurs?
3. What does the kinetic energy depend upon?

Answers

1. Constant rapid motion with random collisions.
2. Elastic collisions.
3. The temperature of the system.

13.2 Gas Pressure

Practice

Questions

Chemistry Chapter 13 States Of Matter Packet Answers

SJ Ball



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