

## Chapter 14 Study Guide

1825 Michael Faraday discovers benzene.

1834 Mitscherlich synthesizes benzene from benzoic acid.

The new compound has an index of hydrogen deficiency(I.H.D.) of 4. IE. 6 carbon structure should have 14 hydrogens( $C_6H_{2n+2}$ ). Benzene only has 6 hydrogens. Therefore, it is missing 8 hydrogens or  $IHD = 4$ . Each index is equal to a double bond or ring. Therefore this is a highly unsaturated compound.

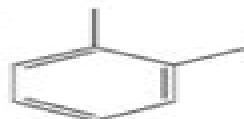
This led to classification of all organics as aliphatic or aromatic.

Kekule def. of aromatic—

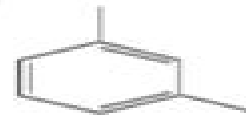
Know structures page 624—

Ortho/para/meta—only refer to disubstituted benzenes.

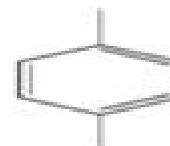
If the two substituents are next to each other it is ortho.



If the two substituents are separated by one carbon it is meta.



If the two substituents are opposite each other it is para.



If both of the substituents are methyl groups the molecule is called xylene.

Nomenclature—substituents are listed alphabetically. If more than two substituents then they are numbered to have the lowest possible number.

If benzene is a substituent it is called phenyl.

The benzyl group is a benzene with a  $CH_2$  attached.

Benzene is non-reactive to ordinary alkene chemical reactions.

# Chemistry Chapter 14 Study Guide

**Arshad Iqbal**



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**Study Guide for Hein and Arena's Foundations of College Chemistry** Peter Scott,Rachael Enriquez Porter,1999-04-18

Study Guide & Solutions Manual to Accompany Organic Chemistry, Third Edition G. Marc Loudon,Joseph G. Stowell,1995

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