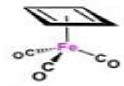
Homework # 1: Due: February 8, 2007

Check the box to the right if you want your graded homework to be placed out in the public rack outside Prof. Stanley's office. Otherwise you will have to pick up your homework from Prof. Stanley in person:

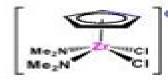
1. (38 pts) Sketch out a structure showing the geometry about the metal center as accurately as possible and clearly show the electron counting for the complexes below. Phosphine ligand abbreviations are defined in your notes (phosphine ligand section). Most molecules (some simplified) are from Organometallics, 2003, # 1 & 2, from the Chem Library or on-line at the Library E-Journal site (only accessable from a campus computer).

a) Fe(η⁴-C₄H₄)(CO)₃ (cover molecule)



Fe(0)	d8
η ⁴ -C ₄ H ₄ 3 CO	4e- 6e-
Total	18e-

C4H_d is cyclobutadiene, a poutral 4o-docor b) cis-[CpZrCl₂(NMe₂)₂]

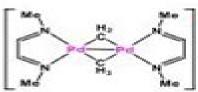


Zr(+4)	dO
Cp	60-
2 NMe ₂	8e-
2.CI	40-
Total	18e-

- c) Pd(NCN)Br (NCN ligand shown below)
 - Pd(+2) d8
 NCN- 6eBr 2e

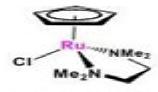
 Total 16e-

d) [Pd₂(μ-CH₃)(μ-CH₂)(N-N)₂]* (one N-N ligand shown below)

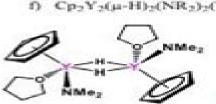


Pd(+2)	d8
N-N	46-
µ-CH ₂ 2-	26-
μ-CH ₃	1e-
Pd-Pd	10-
Total	16e-

e) CpRuCl(Me2NCH2CH2NMe2)



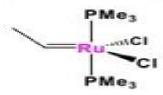
Ru(+2)	d6
Cp-	Ge-
2 NR ₃	46-0-
CI	2e-
Total	18e



THF)2	
Y(+3)	d0
Cp-	6e-
2 µ-H	2e-
THE	20-
NMe _o	40-
Total	14e-
No Y-Y bond	time diversion a

No Y-Y banding due to electron count!

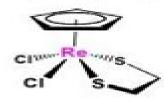
g) Ru(=CHMe)Cl₂(PMe₃)₂



Ru(+4) d4 =CHR² 4e-2 PR₃ 4e-2 Cl 4e-Total 16e-

Also OK to count the ~CHR group as a heutral carbone that is a 2e-donor - you will still get the same total e-count.

h) CpReCl₂(η²-SCH₂CH₂S)



Re(+5)	d2
Cp	6-0-
2 SR-	4e-
2 CI:	40-
Total	16e-

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Padhraic Smyth

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