

NCERT Solutions for Class 11

Chemistry

Chapter – 1 Some basic concepts of Chemistry

1. Calculate the molecular mass of the following:

i. H_2O

Ans: The molecular mass of water, H_2O , can be calculated by the following steps given below:

= (2 \times Atomic mass of hydrogen element) + (1 \times Atomic mass of oxygen element)

= [(2 \times 1.0084 u) + (1 \times 16.00 u)]

= 2.016 u + 16.00 u

= 18.016 u or 18.02 u

ii. CO_2

Ans: The molecular mass of carbon dioxide, CO_2 , is calculated below:

= (1 \times Atomic mass of carbon) + (2 \times Atomic mass of oxygen)

= [(1 \times 12.011 u) + (2 \times 16.00 u)]

= 12.011 u + 32.00 u

= 44.01 u

iii. CH_4

Ans: The molecular mass of methane, CH_4 , is calculated below in step by step manner:

= (1 \times Atomic mass of carbon) + (4 \times Atomic mass of hydrogen)

= [(1 \times 12.011 u) + (4 \times 1.008 u)]

= 12.011 u + 4.032 u

= 16.043 u

2. Calculate the mass percent of different elements present in sodium sulphate (Na_2SO_4).

Ans: The given compound in the question is sodium sulphate and its formula is Na_2SO_4 . Its molecular formula is calculated below:

Chemistry Concepts And Applications Answers Chapter 1

Jeffrey D. Wolt



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Physical Chemistry for JEE Advanced: Part 2, 3E (Free Sample) K. S. Verma, 2022-05-19 Physical Chemistry for JEE Advanced Part 2 a Cengage Exam Crack Series product is designed to help aspiring engineers focus on the subject of physical chemistry from two standpoints To develop their caliber aptitude and attitude for the engineering field and profession To strengthen their grasp and understanding of the concepts of the subjects of study and their applicability at the grassroots level Each book in this series approaches the subject in a very conceptual and coherent manner While its illustrative solved examples facilitate easy mastering of the concepts and their applications an array of solved problems exposes the students to a variety of questions that they can expect in the examination The coverage and features of this series of books make it highly useful for all those preparing for JEE Main and Advanced and aspiring to become engineers

Chemistry Karen Timberlake, 1983 Chemistry John S. Phillips, Cheryl Wistrom, 2000 **The IIT Foundation Series - Physics Class 9, 2/e** , The IIT Foundation Series - Mathematics Class 8, 2/e , *Foundations of Chemistry* Morris Hein, Susan Arena, 1996 A textbook introducing matter atomic theory ionization and other aspects of chemistry to the high school student **Chem C&A Chemla&Min Wksh** McGraw-Hill Education, 1996-08 **Iit Foundations - Physics Class 8** Pearson, 2009-09 IIT Foundation series is specifically for students preparing for IIT right from school days The series include books from class 8 to class 10th in physics chemistry mathematics **Solution Equilibria** F. R. Hartley, Christopher Burgess, R. M. Alcock, 1980 Introduction to General, Organic, and Biochemistry Morris Hein, 1997 The authors recognize that both science and mathematics may be daunting subjects for many students taking this course With this in mind they have anticipated where students might stumble and have paced and organized this text to help them through Their goal is to make the material interesting and relevant so students understand the basic chemical principles related to their career The authors emphasize problem solving and provide a range of practice exercises As in previous editions the text first presents the basic concepts of general chemistry and then moves into organic and biochemistry In this edition the first two sections have been revised primarily to improve explanations and include new pedagogical features The biochemistry portion has been thoroughly updated to include coverage of many recent developments and emerging technologies in the field

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