

Chemistry 101 Final Exam Study Guide- Spring 2017

Acids and Bases

Acid: substance that provides H^+ ions in water

Base: substance that provides OH^- ions in water

Covalent Bonds

- Review: ionic bonds are formed by the positive charge of the cation being attracted to the negative charge of the anion
- Covalent bonds: bonds formed by sharing electrons between the atoms (not ions)
 - 1 electron from each atom (2 atoms per bond)
 - Electrons available for bonding are the un-paired electrons in orbitals (orbitals with only 1 electron in them)
- Molecule
 - A group of atoms held together by covalent bonds
- Atoms will bond with enough other atoms to complete the octet
 - But they can only form as many bonds as unpaired valence electrons
 - Hydrogen has 1 valence electron, so it can only form 1 bond, and will not achieve an octet
 - Boron has 3 valence electrons, so can only form 3 bonds, and will not achieve the octet (only 6 electrons)
 - There are always exceptions to the rules
 - Sulfur and phosphorus can extend the octet

Multiple Covalent Bonds between 2 atoms

- if the octet is not met, additional bonds may be formed
 - CO_2 (carbon dioxide)
 - C has 4 valence electrons, so 4 bonds possible
 - O has 6 valence electrons, 2 are unpaired, so 2 bonds are possible

PRACTICE

IONS:

- Sodium
 - Symbol: Na
 - Valence: 1
 - Electron dot: Na^+
 - Ionic dot: Na^+ (loses its electron to get a 1+ charge)
- Magnesium
 - Symbol: Mg
 - 2 valence e^-
 - Loses 2 electrons to get 2+ charge
- Oxygen
 - Symbol: O
 - Valence: 6
 - Will gain 2 electrons for a 2- charge

Cannot have a covalent bond with ions, just atoms

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N Colangelo



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Digital Learning and Teaching in Chemistry Yehudit Judy Dori, Courtney Ngai, Gabriela Szteinberg, 2023-07-12

Education is always evolving and most recently has shifted to increased online or remote learning Digital Learning and Teaching in Chemistry compiles the established and emerging trends in this field specifically within the context of learning and teaching in chemistry This book shares insights about five major themes best practices for teaching and learning digitally digital learning platforms virtual visualisation and laboratory to promote learning in science digital assessment and building communities of learners and educators The authors are chemistry instructors and researchers from nine countries contributing an international perspective on digital learning and teaching in chemistry While the chapters in this book span a wide variety of topics as a whole they focus on using technology and digital platforms as a method for supporting inclusive and meaningful learning The best practices and recommendations shared by the authors are highly relevant for modern chemistry education as teaching and learning through digital methods is likely to persist Furthermore teaching chemistry digitally has the potential to bring greater equity to the field of chemistry education in terms of who has access to quality learning and this book will contribute to that goal This book will be essential reading for those working in chemical education and teaching Yehudit Judy Dori is internationally recognised formerly Dean of the Faculty of Education of Science and Technology at the Technion Israel Institute of Technology and won the 2020 NARST Distinguished Contributions to Science Education through Research Award DCRA for her exceptional research contributions Courtney Ngai and Gabriela Szteinberg are passionate researchers and practitioners in the education field Courtney Ngai is the Associate Director of the Office of Undergraduate Research and Artistry at Colorado State University Gabriela Szteinberg serves as Assistant Dean and Academic Coordinator for the College of Arts and Sciences at Washington University in St Louis

Resources in

Education ,1996-07 **Iterations, II** Russell Batt, John W. Moore, 1987 *Organic Chemistry Education Research into Practice* Jay Wackerly, Sarah Zingales, Michael Wentzel, Gautam Bhattacharyya, Brett McCollum, 2025-03-25 This Research Topic has three main goals 1 provide a platform for instructors of organic chemistry to showcase evidence based methods and educational theories they have utilized in their classrooms 2 build new and strengthen existing connections between educational researchers and practitioners and 3 highlight how people have used chemical education based research in their teaching practice There are places in the literature dedicated for chemical education research CER however there is not a clear avenue for those that have changed their teaching methods based on published CER and report their experiences Creating this article collection will foster collaboration between chemical education researchers and teachers of organic chemistry This opportunity allows these instructors to share evidence based practices experiences challenges and innovative approaches from CER literature and beyond This Research Topic bridges discipline based education research and the scholarship of teaching and learning which will help advance organic chemistry education and improve student outcomes

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Chemistry Final Exam Study Guide Spring Book Review: Unveiling the Power of Words

In a world driven by information and connectivity, the energy of words has been much more evident than ever. They have the capability to inspire, provoke, and ignite change. Such is the essence of the book **Chemistry Final Exam Study Guide Spring**, a literary masterpiece that delves deep into the significance of words and their effect on our lives. Published by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book's key themes, examine its writing style, and analyze its overall effect on readers.

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