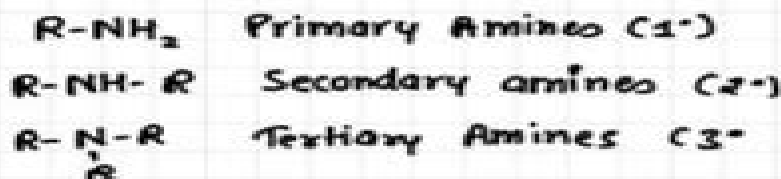


AMINES

Amines are considered as amino of hydrocarbons or alkyl derivatives of ammonia.

Aliphatic amino compounds are called amino alkanes and aromatic amines are called amino arenes eg CH_3NH_2 , $\text{C}_2\text{H}_5\text{NH}_2$, $\text{C}_6\text{H}_5\text{NH}_2$.

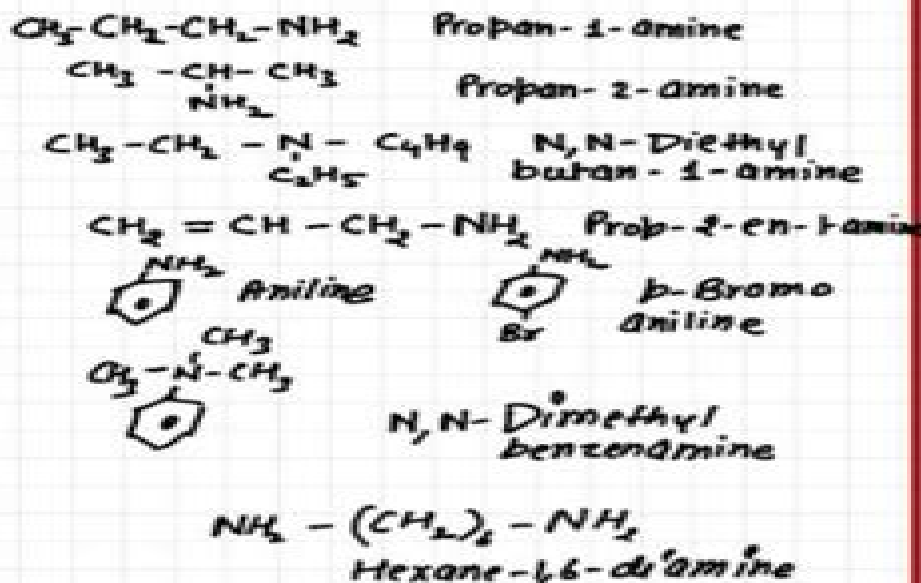


STRUCTURE OF AMINES:



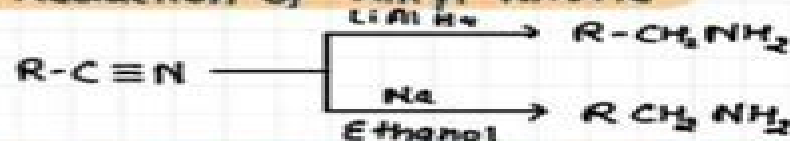
Pyramidal shape of Amine

NOMENCLATURE



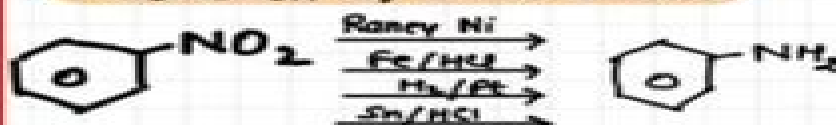
Preparation of Amines:- (24)

► Reduction of Alkyl Nitrile



Rxn is used for ascent of amine

► Reduction of Nitroalkane:



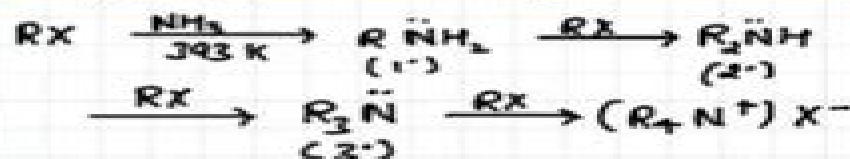
Reduction with Fe/HCl is preferred

► Reduction of Amides:



Reduction of amides give amine with same number of carbons.

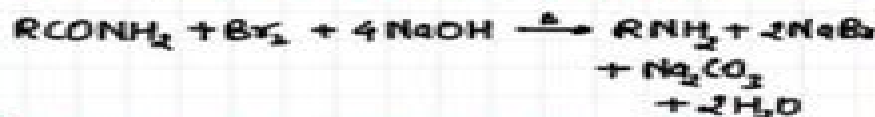
► Hofmann Ammonolysis Method:



Order of reactivity $\rightarrow \text{RI} > \text{RBr} > \text{RCl}$

→ 1° amine is obtained as major product when excess of NH_3 is taken

► HOFMAN BROMAMIDE DEGRADATION:



↳ Amine formed contains one carbon less than that present in the amide.

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NADP/NTN Annual Data Summary, Precipitation Chemistry in the United States ,1989 Bulletin. Library Notes
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