

Balance the equation below



27) Once the equation is balanced, what is the **molar ratio** of water (H<sub>2</sub>O) to oxygen (O<sub>2</sub>)? 1:1

28) If 4.59 grams of O<sub>2</sub> react, how many grams of CH<sub>4</sub> are needed?

$$\frac{4.59 \text{ g-O}_2}{1} \times \frac{1 \text{ mole-O}_2}{32.00 \text{ g-O}_2} \times \frac{1 \text{ mole-CH}_4}{2 \text{ moles-O}_2} \times \frac{16.05 \text{ g CH}_4}{1 \text{ mole-CH}_4}$$

Answer: 1.15 g CH<sub>4</sub>

29) How many grams of CO<sub>2</sub> are needed to react with 8.95 grams of H<sub>2</sub>O?

$$\frac{8.95 \text{ g-H}_2\text{O}}{1} \times \frac{1 \text{ mole-H}_2\text{O}}{18.02 \text{ g-H}_2\text{O}} \times \frac{1 \text{ mole-CO}_2}{2 \text{ moles-H}_2\text{O}} \times \frac{44.01 \text{ g CO}_2}{1 \text{ mole-CO}_2}$$

Answer: 10.93 g CO<sub>2</sub>

Balance the equation below



30) Once the equation is balanced, what is the **molar ratio** of H<sub>2</sub> to O<sub>2</sub>? 2:1

31) How many moles of H<sub>2</sub> are needed to react with 0.42 moles of H<sub>2</sub>O?

$$\frac{0.42 \text{ moles-H}_2\text{O}}{1} \times \frac{2 \text{ moles H}_2}{2 \text{ moles-H}_2\text{O}}$$

Answer: 0.42 moles H<sub>2</sub>

32) How many grams of H<sub>2</sub>O are needed to react with 4.55 moles of H<sub>2</sub>?

$$\frac{4.55 \text{ moles-H}_2}{1} \times \frac{2 \text{ moles-H}_2\text{O}}{2 \text{ moles-H}_2} \times \frac{18.02 \text{ g H}_2\text{O}}{1 \text{ mole-H}_2\text{O}}$$

Answer: 81.99 g H<sub>2</sub>O

33) How many moles of H<sub>2</sub>O are produced when reacted with 8.35 grams of H<sub>2</sub>?

$$\frac{8.35 \text{ g-H}_2}{1} \times \frac{1 \text{ mole-H}_2}{2.02 \text{ g-H}_2} \times \frac{2 \text{ moles H}_2\text{O}}{2 \text{ moles-H}_2}$$

Answer: 4.13 moles H<sub>2</sub>O

# Chemistry Honors Mole Conversion Study Guide

**John M. Dekorte**



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## **Chemistry Honors Mole Conversion Study Guide** Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has been apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Chemistry Honors Mole Conversion Study Guide**," compiled by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve to the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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