

ChemQuest 29

Balancing Equations

Name: _____

Date: _____

Hour: _____

Introduction Questions

1. Write the equation for the reaction of sodium and chlorine (diatomic) to form sodium chloride.
2. Write the equation for the reaction of calcium nitride and sodium chloride to produce calcium chloride and sodium nitride.

Information: Subscripts and Coefficients

A subscript is a small number that tells you how many atoms are in a compound. For example, in CaCl_2 the two is the subscript and it tells us that there are two chloride ions bonded to one calcium.

A coefficient tells also tells us how many atoms or compounds there are, but in a different way. For example in the expression "3 H_2O " the three is the coefficient. The three tells us that there are three molecules of water present. In the expression "3 H_2O " there are a total of 6 hydrogen atoms and 3 oxygen atoms.

Critical Thinking Questions

3. Verify that in $4\text{Ca}_3(\text{PO}_4)_2$ there are 32 oxygen atoms present.
4. How many oxygen atoms are in each of the following:
 _____ a) Al_2O_3 _____ b) 3 Na_2O _____ c) 4 Na_2SO_4 _____ d) 5 $\text{Mg}(\text{NO}_3)_2$

Information: How To Balance Equations

Consider the reaction of sodium and chlorine to produce sodium chloride from gaseous ones:



Remember that when chlorine is by itself it is always written as Cl_2 . On the reactant side of the reaction (left side) there are a total of two atoms of chlorine, but on the product side there is only one atom of chlorine. Atoms cannot simply disappear. In order for the equation to make sense, we need to balance the equation. This can be done, first, by adding a "2" to the product side:



Now the equation reads that one atom of Na reacts with one molecule of Cl_2 to produce two units of NaCl . However, now the Na atoms are not balanced because there is one atom of the reactant side, but two atoms of Na on the product side. This can be fixed by adding another two:



Chemistry Inquiry Chem Quest 29 Answers

David Baud



Chemistry Inquiry Chem Quest 29 Answers:

Reviewing **Chemistry Inquiry Chem Quest 29 Answers**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Chemistry Inquiry Chem Quest 29 Answers**," an enthralling opus penned by a highly acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book's central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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