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Chemistry for Engineers (CHE)
Mastery Test #2

Name: _____ Course and Year: _____ Score: _____
Term: 2nd Semester, 2023-2024 Instructor: Engr. Noel G. Benavides, Ph.D.

Directions: Read each question and their corresponding answers carefully and completely. Choose the answer that best fits the question and write the letter (CAPITALIZED) of your choice on the blank provided before the number.

_____ 1) When the following equation is balanced, the coefficients are _____.
$$\text{NH}_3 (\text{g}) + \text{O}_2 (\text{g}) \rightarrow \text{NO}_2 (\text{g}) + \text{H}_2\text{O} (\text{g})$$

- A) 1, 1, 1, 1 B) 4, 7, 4, 6 C) 2, 3, 2, 3 D) 1, 3, 1, 2 E) 4, 3, 4, 3

_____ 2) When the following equation is balanced, the coefficient of H_2 is _____.
$$\text{K} (\text{s}) + \text{H}_2\text{O} (\text{l}) \rightarrow \text{KOH} (\text{aq}) + \text{H}_2 (\text{g})$$

- A) 1 B) 2 C) 3 D) 4 E) 5

_____ 3) When the following equation is balanced, the coefficient of O_2 is _____.
$$\text{C}_2\text{H}_6\text{O} (\text{g}) + \text{O}_2 (\text{g}) \rightarrow \text{CO}_2 (\text{g}) + \text{H}_2\text{O} (\text{g})$$

- A) 2 B) 3 C) 4 D) 5 E) 1

_____ 4) When the following equation is balanced, the coefficient of H_2 is _____.
$$\text{CO} (\text{g}) + \text{H}_2 (\text{g}) \rightarrow \text{H}_2\text{O} (\text{g}) + \text{CH}_4 (\text{g})$$

- A) 1 B) 2 C) 3 D) 4 E) 0

_____ 5) The balanced equation for the decomposition of sodium azide is _____.

- A) $2\text{NaN}_3 (\text{s}) \rightarrow 2\text{Na} (\text{s}) + 3\text{N}_2 (\text{g})$ B) $2\text{NaN}_3 (\text{s}) \rightarrow \text{Na}_2 (\text{s}) + 3\text{N}_2 (\text{g})$
C) $\text{NaN}_3 (\text{s}) \rightarrow \text{Na} (\text{s}) + \text{N}_2 (\text{g})$ D) $\text{NaN}_3 (\text{s}) \rightarrow \text{Na} (\text{s}) + \text{N}_2 (\text{g}) + \text{N} (\text{g})$
E) $2\text{NaN}_3 (\text{s}) \rightarrow 2\text{Na} (\text{s}) + 2\text{N}_2 (\text{g})$

_____ 6) Of the reactions below, which one is not a combination reaction?

- A) $\text{C} + \text{O}_2 \rightarrow \text{CO}_2$ B) $2\text{Mg} + \text{O}_2 \rightarrow 2\text{MgO}$ C) $2\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$
D) $\text{CaO} + \text{H}_2\text{O} \rightarrow \text{Ca}(\text{OH})_2$ E) $2\text{CH}_4 + 4\text{O}_2 \rightarrow 2\text{CO}_2 + 4\text{H}_2\text{O}$

_____ 7) When a hydrocarbon burns in air, what component of air reacts?

- A) oxygen B) nitrogen C) carbon dioxide D) water E) argon

_____ 8) Of the reactions below, which one is a decomposition reaction?

- A) $\text{NH}_4\text{Cl} \rightarrow \text{NH}_3 + \text{HCl}$ B) $2\text{Mg} + \text{O}_2 \rightarrow 2\text{MgO}$
C) $2\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$ D) $2\text{CH}_4 + 4\text{O}_2 \rightarrow 2\text{CO}_2 + 4\text{H}_2\text{O}$

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