

Serial No. _____

7.) Naturally-occurring copper exists as two isotopes: ^{63}Cu and ^{65}Cu , the masses of which are given below. Which of these isotopes, if either, is most abundant?

| Isotope | Mass (amu) |
|------------------|------------|
| ^{63}Cu | 62.929598 |
| ^{65}Cu | 64.927730 |

- a. More information is needed
 b. Neither, they are nearly-equal abundance
 c. ^{63}Cu is more abundant
 d. ^{65}Cu is more abundant

Handwritten: We can find the answer given the information, and they won't be abundant.

8.) The orbital diagram for the valence shell of a phosphorous atom in its ground state is:

| | | | | |
|----|----------------------|----------------------|----------------------|------------|
| | $\uparrow\downarrow$ | $\uparrow\downarrow$ | \uparrow | |
| a. | $3s$ | | $3p$ | |
| | | \uparrow | \uparrow | \uparrow |
| b. | $3s$ | | $3p$ | |
| | $\uparrow\downarrow$ | | \uparrow | \uparrow |
| c. | $3s$ | | $3p$ | |
| | $\uparrow\uparrow$ | | $\uparrow\downarrow$ | \uparrow |
| d. | $3s$ | | $3p$ | |
| | | $\uparrow\downarrow$ | $\uparrow\downarrow$ | \uparrow |

Section II – Compounds, Reactions, Mole Concept, Solutions, and Stoichiometry

9.) How many moles of xenon tetrafluoride correspond to 1.036 g of this compound?

- a. 2.5 mmol
 b. 1.5 mmol
 c. 500. mmol
 d. 5.0 mmol
 e. None of these

Handwritten: $1.036 \text{ g } \text{XeF}_4 \times \frac{1 \text{ mol}}{354.5 \text{ g}} = 0.0029 \text{ mol} \times 1000 \frac{\text{mmol}}{\text{mol}} = 2.9 \text{ mmol}$

10.)

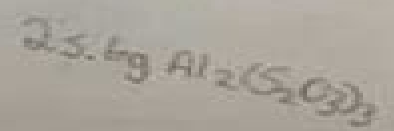
How many iodine atoms are present in 39.0 g of $\text{Ca}(\text{IO}_3)_2$?

- a. 6.002×10^{23}
 b. 3.011×10^{23}
 c. 1.20×10^{24}
 d. 1.20×10^{23}
 e. None of these

Handwritten: $\text{Ca}(\text{IO}_3)_2 \quad 389.88 \text{ g/mol}$
 $39.0 \text{ g} \rightarrow \text{mol} \times 2$



Handwritten: $39.0 \text{ g} \times \frac{1 \text{ mol}}{389.88} \times \frac{2 \text{ mol I}}{1 \text{ mol}} = 0.2 \text{ mol I}$
 $= 0.200$



Handwritten: $4.0 \times 10^{-1} \text{ mol} \times \frac{1 \text{ mol}}{1 \text{ mol}}$

Chemistry Semester 1 Exam Review Answers

Mary Millhollon, Richard H. Langley



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Nursing School Entrance Exam Review William H. Brown, Frederick A. Bettelheim, Jerry March, 2004-04 *Artificial Intelligence in Education. Posters and Late Breaking Results, Workshops and Tutorials, Industry and Innovation Tracks, Practitioners, Doctoral Consortium and Blue Sky* Andrew M. Olney, Irene-Angelica Chounta, Zitao Liu, Olga C. Santos, Ig Ibert Bittencourt, 2024-07-01 This volume constitutes poster papers and late breaking results presented during the 25th International Conference on Artificial Intelligence in Education AIED 2024 which took place in Recife Brazil during July 8 12 2024 The 18 full papers and 92 short papers were carefully reviewed and selected from 200 submissions They are organized in topical sections as follows Part One Blue Sky Industry Innovation and Practitioner Wide AIED and Late Breaking Results Part Two Late Breaking Results Doctoral Consortium Workshops and Tutorials **5 Steps to a 5 AP Chemistry** John Moore, 2003-08-22 For the more than one million students taking the AP exams each year Boxed quotes offering advice from students who have aced the exams and from AP teachers and college professors Sample tests that closely simulate real exams Review material based on the contents of the most recent tests Icons highlighting important facts vocabulary and frequently asked questions Websites and links to valuable online test resources along with author e mail addresses for students with follow up questions Authors who are either AP course instructors or exam developers **The Midland Druggist and Pharmaceutical Review**, 1911 **Resources in Education**, 1985-04 *Midland Druggist and the Pharmaceutical Review*, 1911 The Big Book of Home Learning Volume 1 Getting Started Mary Pride, 2000-09 **Study Guide, Introductory Chemistry, a Foundation ; Introductory Chemistry ; Basic Chemistry** Iris Stovall, Donald J. DeCoste, 2000 **5 Steps to a 5 AP Chemistry, 2015 ed** Richard H. Langley, 2014-07-14 This easy to follow study guide includes a complete course review full length practice tests and access to online quizzes and an AP Planner app 5 Steps to a 5 AP Chemistry features an effective 5 step plan to guide your preparation program and help you build the skills knowledge and test taking confidence you need to succeed This fully revised edition covers the latest course syllabus and matches the new exam It also includes access to McGraw Hill Education's AP Planner app which will enable you to customize your own study schedule on your mobile device AP Planner app features daily practice assignment notifications delivered to your mobile device 2 full length practice AP Chemistry exams Access to online AP Chemistry quizzes 3 separate study plans to fit your learning style **5 Steps to a 5: AP Chemistry 2024** John T. Moore, Richard H. Langley, 2023-07-31 AP Teachers 1 Choice Ready to succeed in your AP course and ace your exam Our 5 Steps to a 5 guides explain the tough stuff offer tons of practice and explanations and help you make the most efficient use of your study time 5 Steps to a 5 AP Chemistry is more than a review guide it's a system that has helped thousands of students walk into test day feeling prepared and confident Everything You Need for a 5 3 full length practice tests that align with the latest College Board requirements Hundreds of practice exercises with answer explanations Comprehensive overview of all test topics Proven strategies from seasoned AP

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