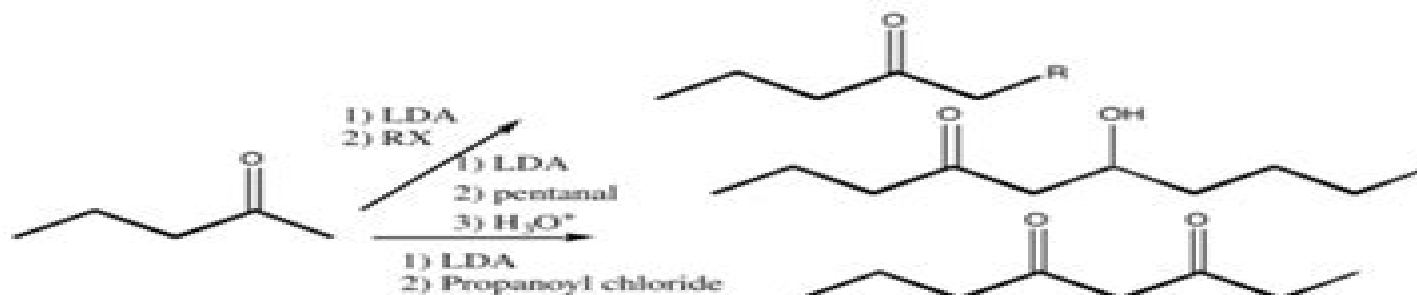


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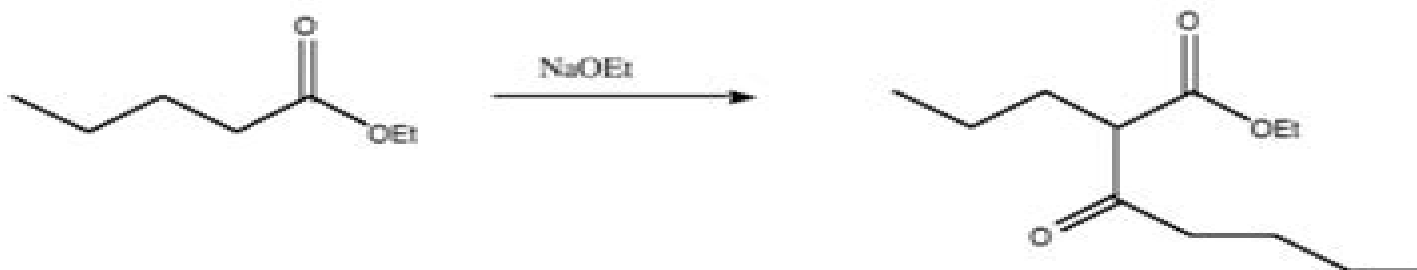
- 1) Enolates can react in a variety of ways.



- 2)  $\beta$ -dicarbonyl compounds are much more acidic than monocarbonyls



- 3) Self "Aldol" reaction with esters — Claisen Condensation. In Chapter 17 we learned that 2 equivalents of an aldehyde or ketone will self-condense in what we know as an aldol reaction. Esters can also do this. You need to use a stronger base. The base must match the ester's alkyl group. You must have 2  $\alpha$ -protons available next to ester.



# Chemistry Chapter 19 Study Guide

**John A. Olmsted, Gregory M. Williams**



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