

Chemical Quantities
Unit Test

Name:

Answer Key.

IB Level

50

1. Answer the following in the space provided. Where required, be sure to use correct scientific notation, significant digits and including proper units of measure. (12 marks)

(a) the product(s) of the reaction of $SO_3(g) + H_2O(l) \rightarrow$ (b) the product(s) of the reaction of $Ba(s) + H_2O(l) \rightarrow$ (c) the product(s) of the reaction of $Na_2CO_3(s) + HNO_3(aq) \rightarrow$ (d) the product(s) of the reaction of $Ag(s) + Pb(NO_3)_2(aq) \rightarrow$ (e) the number of moles in 9.97×10^{25} molecules of CO_2 (f) the number of H atoms in 2.02 moles of $CH_3(CH_2)_8COOH$ (g) the number of moles in a 71.92 g sample of $MgSO_4$ (h) the mass of 3.33×10^{20} molecules of chlorine(i) percentage composition
of ammonium phosphate

chemical formula

molar mass

percentage composition

$H_2SO_4(aq)$	
$Ba(OH)_2(aq) + H_2(g)$	
$NaNO_3(aq) + H_2O(l) + CO_2(g)$	
no rxn.	
166 mol	
2.43×10^{25} H atoms	
0.5974 mol	
0.0392 g	
$(NH_4)_3PO_4$	
149.12 $\frac{g}{mol}$	
28.19% N	8.13% H
20.77% P	42.92% O

Answer the following with complete solutions, be sure to indicate all units of measure and express final answer to correct significant digits.

2. A molecular compound containing the unknown element X has the formula X_4O_{10} . If it is known that the compound is 56.36% oxygen by mass, calculate the molar mass of element X and state its identity. (2 marks)

$$56.36\% = \frac{10(16.00 \frac{g}{mol})}{M_{X_4O_{10}}} \times 100\% \quad 283.9 \frac{g}{mol} = 4(M_X) + 10(16.00 \frac{g}{mol})$$

$$M_X = 30.97 \frac{g}{mol}$$

$$M_{X_4O_{10}} = 283.9 \frac{g}{mol}$$

$\therefore X$ is phosphorus (P)

Chemistry Chapter Chemical Quantities Test Answer Key

Marco Cascella



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Metal-Containing Molecules and Nanomaterials Ana de Bettencourt-Dias, 2024-07-25 Since the discovery and successful deployment of salvarsan to treat syphilis as well as cisplatin as a cancer drug the field of metallodrugs has been flourishing In addition to therapeutic applications metals and metal containing molecules have properties such as an electrochemical or an optical signal that can be used to diagnose biologically relevant molecules or unravel cellular metabolism Moreover metals are part of the essential elements for life and of increased attention in nutraceuticals Most of the current diagnosis processes therapeutics and nutraceuticals are based on organic molecules MILS 26 reviews in 7 comprehensive chapters written by 20 experienced authors the most recent metal containing molecules and nanomaterials for diagnosis and therapy Features The field of metallodrugs has been flourishing and this area of research can be used to diagnose specific biologically relevant molecules or unravel cellular metabolism The authors are preeminent bioinorganic and medicinal inorganic chemists and review the most current research in this field The book discusses the properties and therapeutic applications of metals and metal containing molecules including the use of electrochemical or optical signal for diagnosis of biologically relevant molecules It endorses and stimulates research in the vibrant field of biological inorganic chemistry It reviews the most current research supporting the relevance of metal containing molecules and nanomaterials for diagnosis and therapy

Journal of the Society of Chemical Industry Society of Chemical Industry (Great Britain), 1896

Internal Corrosion of Water Distribution Systems, 2 Edition American Water Works Association, Awwa, 1996-06

Industrial & Engineering Chemistry, 1926

Elementary chemistry William Jerome Harrison, 1890

Quality

Management in the Imaging Sciences E-Book Jeffrey Papp, 2018-09-11 Make sure you have the most up to date quality management information available Quality Management in the Imaging Sciences 6th Edition gives you complete access to both quality management and quality control information for all major imaging modalities This edition includes a new chapter on digital imaging and quality control procedures for electronic image monitors and PACS revisions to the mammography chapter updated legislative content and current ACR accreditation requirements It also features step by step QM procedures complete with full size evaluation forms and instructions on how to evaluate equipment and document results The only text of its kind on the market Papp's is a great tool to help you prepare for the ARRT Advanced Level Examination in Quality Management Special icon identifies federal standards throughout the text alert you to government regulations important to quality management Includes QM for all imaging sciences including fluoroscopy CT MRI sonography and mammography Strong pedagogy aids in comprehension and includes learning objectives chapter outline key terms with definitions in glossary student experiments and review questions at the end of each chapter Step by step QM procedures offer instructions on how to evaluate equipment and full sized sample evaluation forms offer practice in documenting results A practice exam on Evolve includes 200 randomizable practice exam questions for the ARRT advanced certification

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The Chemical News : and Journal of Physical Science ,1907 English Mechanic and Mirror of Science and Art ,1879

Journal of Applied Chemistry ,1963 Vols for 1954 include separately paged section called Abstracts formerly published in British abstracts B I and B II Scientific American ,1877 *The New International Encyclopædia* Frank Moore Colby,Talcott Williams,1918 **The New International Encyclopaedia** Talcott Williams,Frank Moore Colby,1914 *A Treatise on Pharmacy for Students and Pharmacists* Charles Caspari,1901 *The pharmaceutical journal and transactions* ,1893 *The Chemical World* ,1913 Pharmaceutical Journal; ,1893 **New York Review of the Telegraph and Telephone and Electrical Journal** ,1906 Journal of the American Medical Association American Medical Association,1925 *Modern Chemical Technology* Chemical Technician Curriculum Project. Writing Team,1971

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