

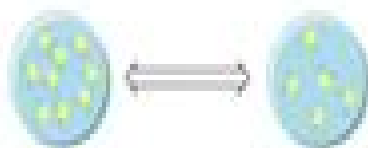
CONCEPT: THE COLLIGATIVE PROPERTIES

- The 4 Colligative Properties discuss what happens to a f solvent as a solute is added to it.
 - As solute is added to a solvent some colligative properties ____ and others ____.
 - ____ Boiling Point and Osmotic Pressure.
 - ____ Freezing Point and Vapor Pressure.

Colligative Properties

Boiling Point (BP)

Temperature where a ____ & ____ are in equilibrium.



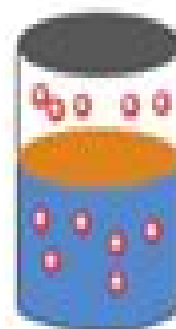
Freezing Point (FP)

Temperature where a ____ & ____ are in equilibrium.



Vapor Pressure

The pressure exerted by a ____ at the surface of a liquid.
□ Measureable at equilibrium.



Osmotic Pressure

The force that drives osmosis from ____ concentration to ____ concentration.



Chemistry Chapter Colligative Properties Quiz Key

R Pring



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In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is really astonishing. Within the pages of "**Chemistry Chapter Colligative Properties Quiz Key**," an enthralling opus penned by a highly acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve in to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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