Question 2

- 2(b) The zinc decreases in size, the blue colour of the solution fades and a red-brown solid is deposited in the test tube and/or on the solid zinc.
- 2(c) Add the suspected reducing agent to a few cm³ of an oxidizing agent such as H*_(ac)/KMnO_{4(ac)} in a test tube. If the KMnO⁴ changes from purple to colourless, the presence of a reducing agent is confirmed.

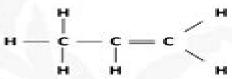
Question 3

- 3(a) Members of a homologous series:
 - Have the same general formula.
 - Show similar chemical reactions though varying in vigour.
 - Differ in molecular formula from the next by CH₂.
- 3(b)(ii)



Compound A

- 3(b)(ii) Compound A belongs to the alkanes.
- 3(b)(iii) C₅H₁₂.
- 3(c)(i)



Compound B

- 3(c)(ii) Compound B belongs to the alkenes.
- 3(c)(iii) C₅H₁₀-
- 3(d)(i) Carboxylic acid: $C_nH_{2n-1}COOH$ where $n \ge 0$.
 - Alcohol: C,H2m-1OH where n ≥ 1.
- 3(d)(iii) Carboxylic acid: methanoic acid.

Alcohol: methanol.

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