

## Question 2

- 2(b) The zinc decreases in size, the blue colour of the solution fades and a red-brown solid is deposited in the test tube and/or on the solid zinc.
- 2(c) Add the suspected reducing agent to a few cm<sup>3</sup> of an oxidizing agent such as H<sup>+</sup><sub>(aq)</sub>/KMnO<sub>4(aq)</sub> in a test tube. If the KMnO<sub>4</sub> changes from purple to colourless, the presence of a reducing agent is confirmed.

## Question 3

- 3(a) Members of a homologous series:
- Have the same general formula.
  - Show similar chemical reactions though varying in vigour.
  - Differ in molecular formula from the next by CH<sub>2</sub>.

3(b)(i)



Compound A

3(b)(ii) Compound A belongs to the alkanes.

3(b)(iii) C<sub>3</sub>H<sub>12</sub>

3(c)(i)



Compound B

3(c)(ii) Compound B belongs to the alkenes.

3(c)(iii) C<sub>5</sub>H<sub>10</sub>

3(d)(i) Carboxylic acid: C<sub>n</sub>H<sub>2n+1</sub>COOH where n ≥ 0.

Alcohol: C<sub>n</sub>H<sub>2n+1</sub>OH where n ≥ 1.

3(d)(ii) Carboxylic acid: methanoic acid.

Alcohol: methanol.

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