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THIRD TERM EXAMINATION

SUBJECT: CHEMISTRY

CLASS: S. S. S ONE

DURATION: 2HR

1. Acid turns blue litmus paper _____?
a. yellow b. colourless c. green d. red
2. Acids are characterized by a. Bitter taste b. soapy taste
c. sour taste d. all listed
3. An element existing in two or more physical states and maintaining the same chemical characteristics is called _____ a. polymerization b. allotropes
c. isotopes d. crystal
4. Which allotrope of carbon is used in lead pencil?
a. graphite b. coke c. coal d. diamond
5. Carbon dioxide is a gas that burns with _____ flame
a. yellow b. green c. blue d. colourless
6. Hydrocarbons are of two classes namely: a. graphite and diamond
b. coke and coal c. aliphatic and gas d. aromatic and aliphatic
7. An example of aromatic hydrocarbons is a. Ethane b. methane
c. cyclopropane d. benzene
8. Which is not a raw material for chemical industry a. petroleum
b. coal c. alum d. salt
9. Which is not a use of hydrogen trioxocarbonate a. As baking powder
b. As health salts b. As fire extinguishers d. To manufacture hydroxide
10. Which is not a property of carbon(II)oxide a. It is colourless
b. It is slightly lighter than air c. It is insoluble in water d. It is soluble
11. Which is a type of amorphous a. carbon black charcoal b. plant charcoal
c. red charcoal d. graded charcoal
12. Which of these is not a type of coal? A. peat b. lignite bituminous c. coal
13. Coke is a non-volatile residue, this statement is a. true b. false c. negative
14. Water gas is a mixture that contains CO and _____ gas? a. hydrogen
b. helium c. lithium d. berlium
15. The number of replaceable hydrogen atom in one molecule of an acid indicates its
a. Basicity b. Alkalinity c. Acidity d. Reactivity
16. Which of the following is /are characteristics properties of acids? They i. form
hydroxonium ions in water ii. liberate carbon (iv)oxide from trioxocarbonate(iv) salts. iii.
Turn blue litmus paper red. iv. have a soapy feel
a. iv only b. i and ii c. iii and iv d. i, ii, and iii
17. The following acid are monobasic except a. Trioxonitrate(v)acid b. Hydrochloric acid
c. Ethanoic acid d. Tetraoxosulphate (v)acid
18. Which of the following is an acid salt? a. $(\text{NH}_4)_2\text{CO}_3$ b. CH_3COONa c. KHSO_4 d. MgSO_4
19. Alums are classified as a. basic salts b. acid salts c. anhydrous salts d. double salts
20. Which of the following conclusion about solution of pH4 is correct
a. it contains more OH^- than H_3O^+ b. its POH value will be 10 c. it is more acidic than a
solution of pH2

Chemistry Exams For Third Term Ss 1 2015 Session

Xiaolong Qi



Chemistry Exams For Third Term Ss 1 2015 Session:

This book delves into Chemistry Exams For Third Term Ss 1 2015 Session. Chemistry Exams For Third Term Ss 1 2015 Session is a crucial topic that must be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Chemistry Exams For Third Term Ss 1 2015 Session, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:
 - Chapter 1: Introduction to Chemistry Exams For Third Term Ss 1 2015 Session
 - Chapter 2: Essential Elements of Chemistry Exams For Third Term Ss 1 2015 Session
 - Chapter 3: Chemistry Exams For Third Term Ss 1 2015 Session in Everyday Life
 - Chapter 4: Chemistry Exams For Third Term Ss 1 2015 Session in Specific Contexts
 - Chapter 5: Conclusion
 2. In chapter 1, the author will provide an overview of Chemistry Exams For Third Term Ss 1 2015 Session. The first chapter will explore what Chemistry Exams For Third Term Ss 1 2015 Session is, why Chemistry Exams For Third Term Ss 1 2015 Session is vital, and how to effectively learn about Chemistry Exams For Third Term Ss 1 2015 Session.
 3. In chapter 2, this book will delve into the foundational concepts of Chemistry Exams For Third Term Ss 1 2015 Session. The second chapter will elucidate the essential principles that must be understood to grasp Chemistry Exams For Third Term Ss 1 2015 Session in its entirety.
 4. In chapter 3, the author will examine the practical applications of Chemistry Exams For Third Term Ss 1 2015 Session in daily life. This chapter will showcase real-world examples of how Chemistry Exams For Third Term Ss 1 2015 Session can be effectively utilized in everyday scenarios.
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 6. In chapter 5, this book will draw a conclusion about Chemistry Exams For Third Term Ss 1 2015 Session. This chapter will summarize the key points that have been discussed throughout the book.
- This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Chemistry Exams For Third Term Ss 1 2015 Session.

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