

Final Exam Ver. 8 (90 Multiple Choice)

Multiple Choice: (90 marks)

Circle the letter of the choice that best completes the statement or answers the question.

1. Which of the following is an ester?



A) none of the above B) 1 C) 2 D) 3 E) 4

2. A neutral fluorine atom has a much greater attraction for electrons than does any other neutral atoms. Consider the following series of molecules:

I SiH_4 , II CH_3F III CH_3F_2 IV CHF_3 V CF_4

Which of the above molecules will have a dipole?

A) All of the above B) Molecules II and IV only. C) Molecules II, III and IV only.
 D) Molecules II, III and V only. E) None of the above

3. Given the structural formulas:



Which two formulas represent compounds that are isomers of each other?

A) A and B B) B and D C) A and C D) C and D

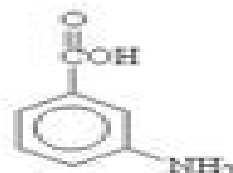
4. The IUPAC name for the compound



is

A) 2-methylpentanamine. B) 4-methylpentanamine.
 C) 2-methyl-4-pentanamine. D) 4-methyl-2-pentanamine.

5. The compound below is named



A) benzoic acid amine B) 3-aminobenzoic acid C) 2-acid aniline
 D) benzamide E) 2-aminobenzoic acid

Chemistry Final Exam Mc Answers

R Sandford



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