

Review Sheet for Chemistry First Semester Final

Refer to your class notes, worksheets, and the textbook to complete this review sheet. Study early so that you will have time to ask questions about what you don't understand. For the exam, you will need to bring 2 number 2 pencils, a scientific calculator (there will be NO sharing and NO cell phones for calculations) and you may bring a 4x6 card with formulas, equations etc that you feel you need. This will be collected with your test and must have your name on it. (MUST be a 4x6 CARD, no paper cut to 4x6- MUST be HAND written)

Things you need to know: In addition to the sections below:

SI Base Units- Pg 73

Metric Prefixes- Page 74

Know how to convert from one unit to another (ie: kilometers to millimeters)

KNOW YOUR ELEMENTS – The ones you were tested on! (name, symbol and spelling counts!)

KNOW the Polyatomic Ions on Pg 257 (charge, name, formula)

KNOW the Common Metal Ions with more than one charge (table on page 255)

KNOW the Prefixes for Covalent molecule names (page 269)

BE ABLE TO USE the Significant Figures rules

Ch. 1: Introduction to Chemistry

manipulated (independent) variable

responding (dependent) variable

observation

theory

steps in scientific method

problem solving in chemistry

Ch. 2: Matter and Change

chemical change

chemical property

chemical reaction

compound

element

filtration

gas

heterogeneous mixture

homogeneous mixture

conservation of mass

liquid

mass

mixture

phase

physical change

physical property

product

reactant

solid

solution

substance

vapor

volume

Properties of Matter

Mixtures

Elements and Compounds

Chemical Reactions

Matter: Anything that takes up space and has mass

➤ Physical Changes and Chemical Changes

Define each. How can you tell the difference between the two?

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ML Morrison



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