

ANSWERS TO "TRY YOURSELF" PROBLEMS FROM STUDY SECTION 2.5

Try Yourself 2.5

Write formulas for all the ionic compounds that can be formed by combining the following ions:



Answers:

$(\text{NH}_4)_2\text{CO}_3$ Ammoniumcarbonaat. / *Ammonium carbonate.*

$(\text{NH}_4)_3\text{PO}_4$ Ammoniumfosfaat. / *Ammonium phosphate.*

NiCO_3 Nikkel(II)carbonaat. / *Nickel(II) carbonate.*

$\text{Ni}_3(\text{PO}_4)_2$ Nikkel(II)fosfaat. / *Nickel(II) phosphate.*

Try Yourself 2.6

Which compound formula and name in the list is NOT correct?

- | | | |
|--|--|--------------------------------------|
| 1. CaSO_4 , calcium sulfate | 2. NaNO_3 , sodium nitrate | 3. MgI_2 , magnesium iodide |
| 4. NH_4PO_4 , ammonium phosphate | 5. $\text{Ca}(\text{ClO})_2$, calcium hypo-chlorite | |

Answers:

- | | |
|---|---|
| 1. CaSO_4 , calcium sulfate | (Correct) |
| 2. NaNO_3 , sodium nitrate | (Correct) |
| 3. MgI_2 , magnesium iodide | (Correct) |
| 4. NH_4PO_4, ammonium phosphate | (Incorrect, should be $(\text{NH}_4)_3\text{PO}_4$) |
| 5. $\text{Ca}(\text{ClO})_2$, calcium hypo-chlorite | (Correct) |

Try Yourself 2.7

Sodium oxalate has the formula $\text{Na}_2\text{C}_2\text{O}_4$. Based on this information, what is the formula for iron(III) oxalate?

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