

- $$\text{H}:\ddot{\text{O}}:\ddot{\text{O}}:\text{H} \quad \text{or} \quad \begin{array}{cc} :\ddot{\text{O}}: & :\ddot{\text{O}}: \\ \text{H} & \text{H} \end{array}$$

- | Molecule: | AlH_3 | SiF_4 | NH_3 | H_2Te | HCl |
|--------------------|------------------------------------|---|-----------------------------------|------------------------------------|----------------------------------|
| Lewis Structure: | $\text{H}:\text{Al}:\text{H}$
H | $\begin{array}{c} \cdot\cdot \\ \cdot\text{F} \\ \cdot\cdot \\ \cdot\text{F}:\text{Si}:\text{F} \\ \cdot\cdot \\ \cdot\text{F} \\ \cdot\cdot \end{array}$ | $\text{H}:\text{N}:\text{H}$
H | $\text{H}:\text{Te}:\text{H}$
H | $\text{H}:\text{Cl}:$
central |
| Steric #: | 3 | 4 | 4 | 4 | 4 |
| Shape: | Δ PLANAR | TETRA | Δ PYRAMID. | BENT | LINEAR |
| Polar / Nonpolar?: | NON | NON | POLAR | POLAR | POLAR |

- | Molecule: | AlH_3 | HF | NH_3 | H_2S | Ar |
|-----------|---|---|---|---|---|
| IMF: | London dispersion
Dipole-dipole
H-bonding | London dispersion
Dipole-dipole
H-bonding | London dispersion
Dipole-dipole
H-bonding | London dispersion
Dipole-dipole
H-bonding | London dispersion
Dipole-dipole
H-bonding |

Electronegativity Values

Electronegativity Values																		NON POLAR MOLECULES				
1 H 2.1													5 B 2.0	6 C 2.5	7 N 3.0	8 O 3.5	9 F 4.0					
3 Li 1.0	4 Be 1.5											13 Al 1.5	14 Si 1.8	15 P 2.1	16 S 2.5	17 Cl 3.0						
11 Na 1.0	12 Mg 1.2											31 Ga 1.7	32 Ge 1.9	33 As 2.1	34 Se 2.4	35 Br 2.8						
19 K 0.9	20 Ca 1.0	21 Sc 1.3	22 Ti 1.4	23 V 1.5	24 Cr 1.6	25 Mn 1.6	26 Fe 1.7	27 Co 1.7	28 Ni 1.8	29 Cu 1.8	30 Zn 1.6	49 In 1.6	50 Sn 1.8	51 Sb 1.9	52 Te 2.1	53 I 2.5						
37 Rb 0.9	38 Sr 1.0	39 Y 1.2	40 Zr 1.3	41 Nb 1.5	42 Mo 1.6	43 Tc 1.7	44 Ru 1.8	45 Rh 1.8	46 Pd 1.8	47 Ag 1.6	48 Cd 1.6	81 Tl 1.6	82 Pb 1.7	83 Bi 1.8	84 Po 1.9	85 At 2.1						
55 Cs 0.8	56 Ba 1.0	57 La 1.1	72 Hf 1.3	73 Ta 1.4	74 W 1.5	75 Re 1.7	76 Os 1.9	77 Ir 1.9	78 Pt 1.8	79 Au 1.9	80 Hg 1.7											
87 Fr 0.8	88 Ra 1.0	89 Ac 1.1																				

Chemistry If8766 Page 48

Sebastian Brünink

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Chemistry If8766 Page 48 Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has are more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Chemistry If8766 Page 48**," published by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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