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CHEMISTRY, PAPER-II

PART-I(M	OWED: THREE HOURS CQS): MAXIMUM 30 MINUTES	PART-I (MCQS) MAXIMUM MARKS = 20 PART-II MAXIMUM MARKS = 80
(ii) (iii) (iv) (v)	All the parts (if any) of each Question must Candidate must write Q. No. in the Answer No Page/Space be left blank between the crossed.	RT-II. ALL questions carry EQUAL marks. be attempted at one place instead of at different places. Book in accordance with Q. No. in the Q.Paper. answers. All the blank pages of Answer Book must be
(vi)	Extra attempt of any question or any part of the attempted question will not be considered.	
(vii)	Use of calculator is allowed.	

- PART-II Arrange the following alkenes in order of their relative stability. How will you proceed to (a) (5) determine the order practically? 1-hexene i. ii. cis-3-hexene
 - Bill. trans-3-hexene (iv) 2-methyl-2-pentene (*) 2,3-dimethyl-2-butene
 - (b) Explain why? Poly substitution is a complicating factor in aromatic alkylation but not in **(:)**
 - aromatic nitration or halogenation. (iii) A undergoes nitration predominantly at the ortho/ para positions but B mainly at meta position

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- Compare the basicity of: (CCl₃)₃N
 - (CH₃)₃N (i) (iii) C6H5CH2NH2
 - CH3C6H4NH2 80 (iiii) Aniline Cyclohexyl amine

- Explain why? (2.5 each) (d) Tertiary carbocation is more stable than primary.
- (ii) Ethanol has higher boiling point than diethyl ether. NEW PASTURY COR WWW.FRETUENCOM Write the structural formula for more stable conformation of each of the following
- Q. 3. (a) compounds.
 - a) trans-1-Fluoro-3-methylcyclohexane,
 - b) cis-1-lodo-4-methylcyclohexane
 - c) cis-1-tert-Butyl-4-methylcyclohexane, ___ WWW.PSSTUDY.COM d) cis-1,3,5-Trimethylcyclohexane

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