

	I	II	III			
Final burette reading	17.45	32.90	36.05			
Initial burette reading	2.10	17.45	20.60			
Volume of solution B used (cm ³)	15.35	15.45	15.45			
Test tube number	1	2	3	4	5	6
Volume of distilled water (cm ³)	0	2	3	5	6	7
Volume of solution A (cm ³)	10	8	7	5	4	3
Time (s)	22.5	28.0	32.0	50.0	57.5	85.0
Rate = V_{max}/t (s ⁻¹)	0.044	0.036	0.031	0.020	0.017	0.012

(4 marks)

(a) - (i) Average volume

$$= 15.35 + 15.45 + 15.45$$

$$= 15.42 \text{ cm}^3$$

(1 mark)

(ii) Moles of sodium thiosulphate used

$$= \frac{0.05 \times 15.42}{1000} = 7.71 \times 10^{-4} \text{ moles} \quad (2)$$

(1 mark)

(b) (i) Number of moles of A in 25.0 cm³



$$7.71 \times 10^{-4} / 6 = 1.28 \times 10^{-4} \text{ moles} \quad (1)$$

(1 mark)

solution A in mol dm³

$$\frac{1.28 \times 10^{-4} \text{ moles in } 25\text{-cm}^3}{7 \text{ moles in } 1000\text{-cm}^3}$$

$$\frac{1.28 \times 10^{-4} \times 1000}{25} (1)$$

$$5.12 \times 10^{-3} \text{ moles/dm}^3 (1)$$

(2 marks)

Table 2

(1) (1) (1) (1) (1) (1)

(6 marks)

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