

Diagram of a hydrogen atom. The nucleus contains one proton (positive charge) and one neutron (neutral charge). The electron cloud surrounds the nucleus and contains one electron (negative charge). The label "Hydrogen Atom" is placed within the electron cloud.

Reaction formulae

$\text{NaCl} + \text{H}_2\text{O} \rightarrow \text{NaOH} + \text{HCl}$

The number of acid atoms and basic atoms reacting are the same. 1:1 molar ratio.

How to calculate the molar of NaOH used:

If a reaction of the reactants are present in the same ratio i.e. NaCl to NaOH

The molar will be 100g / 40g

The molar concentration = 100g / 40g

Unit 1 Chemistry

Displacement reactions always involve the most reactive metal ending up bonded whilst the least reactive is unbonded.

1) Copper sulfate + Magnesium → Magnesium sulfate + Copper

2) Gold chloride + Zinc → Zinc chloride + Gold

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