

STOICHIOMETRY WORKSHEET

1. Calculate the number of moles of NaOH that are needed to react with 500 g of H_2SO_4 according to the following equation:



2. Calculate the mass of NH_3 that can be produced from the reaction of 125 g of NCl_3 according to the following equation:



3. Identify the limiting reactant and determine the mass of CO_2 that can be produced from the reaction of 25 g of C_3H_8 with 75 g of O_2 according to the following equation:



4. How many grams of SO_2 are produced when 152 g of CS_2 reacts with 48.0 g of O_2 according to the following equation:



Chemistry Stoichiometry Problems 1 Answers

Lei Shi



Chemistry Stoichiometry Problems 1 Answers:

Whispering the Techniques of Language: An Psychological Journey through **Chemistry Stoichiometry Problems 1 Answers**

In a digitally-driven earth wherever screens reign great and quick conversation drowns out the subtleties of language, the profound secrets and mental subtleties hidden within words frequently move unheard. However, situated within the pages of **Chemistry Stoichiometry Problems 1 Answers** a charming fictional value pulsating with organic feelings, lies an extraordinary quest waiting to be undertaken. Published by an experienced wordsmith, that wonderful opus attracts readers on an introspective trip, softly unraveling the veiled truths and profound affect resonating within the fabric of each word. Within the mental depths of this moving review, we can embark upon a sincere exploration of the book is core styles, dissect its captivating writing fashion, and yield to the strong resonance it evokes deep within the recesses of readers hearts.

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