

Name _____

Date _____

Chemical Reactions

Matching (1 each)

1. _____ This speeds up a chemical reaction that takes too long to happen.
2. _____ When a solid is formed during a chemical change.
3. _____ The process where one or more substances change into another.
4. _____ The minimum amount of energy needed to start a chemical reaction.
5. _____ This is a substance that slows down a chemical reaction.
6. _____ The final result of a chemical reaction.
7. _____ When energy is put into a chemical to create a reaction.
8. _____ The force that holds 2 atoms together.
9. _____ When energy is released from a chemical reaction.
10. _____ Substances that are used in a chemical reaction.

a.	Precipitate	f. Endothermic Reaction
b.	Activation Energy	g. Exothermic Reaction
c.	Chemical Reaction	h. Product
d.	Catalyst	i. Reactant
e.	Inhibitor	j. Bond

True and False (2 each)

11. _____ The following equation is balanced - $2\text{NaCl} + \text{F}_2 \rightarrow 2\text{NaF} + \text{Cl}_2$
12. _____ Bonds break when the molecules hit each other hard enough.
13. _____ The following equation is exothermic - $2\text{H}_2\text{O} + \text{energy} \rightarrow 2\text{H}_2 + \text{O}_2$
14. _____ The higher a temperature the faster a chemical reaction takes place.
15. _____ The less surface area the faster a chemical reaction will take place.

Multiple Choice (2 each)

16. _____ Which of the following is NOT a sign of a chemical reaction?
a. Gas Formation b. Energy Change c. Chemical Loss d. Color Change
17. _____ What type of reaction is the following equation? - $2\text{NaCl} + \text{F}_2 \rightarrow 2\text{NaF} + \text{Cl}_2$
a. Single-Displacement b. Double-Displacement c. Synthesis d. Decomposition
18. _____ $\text{H}_2\text{CO}_3 \rightarrow \text{H}_2\text{O} + \text{CO}_2$
a. Single-Displacement b. Double-Displacement c. Synthesis d. Decomposition
19. _____ How many atoms are in the equation? - 2NaCl_2
a. 5 b. 1 c. 6 d. 3
20. _____ How many atoms are in the equation? - $\text{C}_2\text{H}_5\text{O}_2$
a. 24 b. 12 c. 9 d. 15

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N Colangelo



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Table of Contents Chemistry Ch Assessment Chemical Reactions

1. Understanding the eBook Chemistry Ch Assessment Chemical Reactions
 - The Rise of Digital Reading Chemistry Ch Assessment Chemical Reactions
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Ch Assessment Chemical Reactions
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Ch Assessment Chemical Reactions
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Ch Assessment Chemical Reactions
 - Personalized Recommendations
 - Chemistry Ch Assessment Chemical Reactions User Reviews and Ratings
 - Chemistry Ch Assessment Chemical Reactions and Bestseller Lists
5. Accessing Chemistry Ch Assessment Chemical Reactions Free and Paid eBooks
 - Chemistry Ch Assessment Chemical Reactions Public Domain eBooks
 - Chemistry Ch Assessment Chemical Reactions eBook Subscription Services
 - Chemistry Ch Assessment Chemical Reactions Budget-Friendly Options

6. Navigating Chemistry Ch Assessment Chemical Reactions eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemistry Ch Assessment Chemical Reactions Compatibility with Devices
 - Chemistry Ch Assessment Chemical Reactions Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry Ch Assessment Chemical Reactions
 - Highlighting and Note-Taking Chemistry Ch Assessment Chemical Reactions
 - Interactive Elements Chemistry Ch Assessment Chemical Reactions
8. Staying Engaged with Chemistry Ch Assessment Chemical Reactions
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Ch Assessment Chemical Reactions
9. Balancing eBooks and Physical Books Chemistry Ch Assessment Chemical Reactions
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Ch Assessment Chemical Reactions
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chemistry Ch Assessment Chemical Reactions
 - Setting Reading Goals Chemistry Ch Assessment Chemical Reactions
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemistry Ch Assessment Chemical Reactions
 - Fact-Checking eBook Content of Chemistry Ch Assessment Chemical Reactions
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

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