

# Chemistry 105 Equation Sheet

## Constants

$$\begin{aligned}a_0 &= 0.5291772059 \text{ \AA} \\c &= 2.99792458 \times 10^8 \text{ m/s} \\e &= 1.6021765 \times 10^{-19} \text{ C} \\\epsilon_0 &= 8.8541877 \times 10^{-12} \text{ F/m} \\\hbar &= 6.626 \times 10^{-34} \text{ J}\cdot\text{s} \\k_B &= 1.38065 \times 10^{-23} \text{ J/K} \\k_B &= 8.987551 \times 10^8 \frac{\text{J}\cdot\text{m}^2}{\text{mol}\cdot\text{K}\cdot\text{C}^2} \\m_e &= 9.109383 \times 10^{-31} \text{ kg} \\m_n &= 1.6749273 \times 10^{-27} \text{ kg} \\m_p &= 1.6726217 \times 10^{-27} \text{ kg} \\N_A &= 6.022142 \times 10^{23} \text{ mol}^{-1} \\R &= 8.31447 \frac{\text{J}}{\text{mol}\cdot\text{K}} \\&= 0.0820575 \frac{\text{L}\cdot\text{atm}}{\text{mol}\cdot\text{K}} \\&= 1.987207 \frac{\text{cal}}{\text{mol}\cdot\text{K}} \\R_H &= 1.0974 \times 10^7 \text{ m}^{-1}\end{aligned}$$

## Conversions (\* = exact)

### Length

$$\begin{aligned}1 \text{ in} &= 2.54 \text{ cm}^* \\1 \text{ ft} &= 12 \text{ inches}^* \\1 \text{ yd} &= 3 \text{ ft}^* \\1 \text{ m} &= 39.37 \text{ inches}^* \\1 \text{ mile} &= 5280 \text{ feet}^* \\1 \text{ mile} &= 1.609 \text{ km}\end{aligned}$$

### Mass

$$\begin{aligned}1 \text{ lb} &= 453.59237 \text{ g} \\&= 16 \text{ oz. avdp.}^* \\1 \text{ US lb} &= 12 \text{ troy ounces}^* \\1 \text{ ton} &= 2000 \text{ lb}^* \\1 \text{ amu} &= 1.660538 \times 10^{-24} \text{ g}\end{aligned}$$

### Volume

$$\begin{aligned}1 \text{ US gal} &= 3.7854 \text{ L} = 4 \text{ US qt}^* \\1 \text{ US qt} &= 32 \text{ US fl. oz.}^* \\1 \text{ US fluid oz.} &= 29.5735 \text{ mL} \\1 \text{ cm}^3 &= 0.001 \text{ L}^* \\1 \text{ in}^3 &= 16.4 \text{ cm}^3\end{aligned}$$

## Pressure

$$\begin{aligned}1 \text{ atm} &= 760 \text{ torr}^* \\&= 101325 \text{ Pa}^* \\&= 406.783 \text{ in H}_2\text{O} \\&= 14.6959 \text{ lb/in}^2 \text{ (psi)} \\1 \text{ torr} &= 1 \text{ mm Hg (at } 0^\circ\text{C)}^* \\1 \text{ Pa} &= 1 \text{ kg}\cdot\text{m}^{-1}\cdot\text{s}^{-2} \\1 \text{ bar} &= 100,000 \text{ Pa}^*\end{aligned}$$

## Force

$$1 \text{ N} = 1 \text{ kg}\cdot\text{m}\cdot\text{s}^{-2}$$

## Power

$$1 \text{ Watt} = 1 \text{ J/s}$$

## Energy

$$\begin{aligned}1 \text{ J} &= 1 \text{ kg}\cdot\text{m}^2/\text{s}^2 = 0.2390\text{-cal} \\1 \text{ J} &= 1 \text{ V}\cdot\text{C} = 0.009369 \text{ L}\cdot\text{atm} \\1 \text{ eV} &= 1.602 \times 10^{-19} \text{ J} \\1 \text{ L}\cdot\text{atm} &= 101.3 \text{ J}\end{aligned}$$

## Formulas

$$\begin{aligned}\rho &= m/V \\T_F &= \frac{9}{5}T_C + 32^\circ\text{F} \\T_K &= \frac{1}{1}T_C + 273.15 \text{ K} \\c &= \lambda \nu \\E &= h\nu \\\lambda &= \frac{h}{mv} \\E_{\text{kinetic}} &= \frac{1}{2}mv^2 \\E &= mc^2 \\\Phi &= h\nu - KE_{\text{electron}} \\E_n &= -Z^2 \frac{R_H h c}{n^2}\end{aligned}$$

$$\Delta x \Delta(mv) \geq \frac{h}{4\pi}$$

$$\frac{1}{\lambda} = R_H \left( \frac{1}{n_1^2} - \frac{1}{n_2^2} \right)$$

$$V(r) = \frac{q_1 q_2}{4\pi\epsilon_0 r}$$

$$F(r) = k_e \frac{Q_1 Q_2}{r^2}$$

$$\mu = Qr$$

$$\Delta E = q + w$$

$$\Delta E_{\text{univ}} = 0$$

$$\Delta S_{\text{univ}} \geq 0$$

$$S = k_B \ln W$$

$$q = mC\Delta T$$

$$w = -P_{\text{ext}} \Delta V$$

$$\Delta H = \Delta E + P\Delta V$$

$$\Delta S = \frac{q_{\text{rev}}}{T}$$

$$\Delta G = \Delta H - T\Delta S$$

$$\Delta G = \Delta G^\circ + RT \ln Q$$

$$\Delta H'_{\text{rxn}} = \sum nH'_{f,\text{prod}} - \sum nH'_{f,\text{react}}$$

$$\Delta H'_{\text{rxn}} \approx \sum nD_{\text{broken}} - \sum nD_{\text{formed}}$$

$$PV = nRT$$

$$P_{\text{tot}} = P_1 + P_2 + P_3 + \dots$$

$$\frac{\text{Rate}_A}{\text{Rate}_B} = \frac{\text{Time}_B}{\text{Time}_A} = \sqrt{\frac{M_B}{M_A}}$$

$$KE_{\text{ave}} = \frac{3}{2}RT$$

$$u_{\text{rms}} = \sqrt{v^2} = \sqrt{\frac{3RT}{M}}$$

$$\text{pX} = -\log X$$

$$K_w = [\text{H}_3\text{O}^+][\text{OH}^-]$$

\* = exact  $\nu$  = frequency  $v$  = velocity

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# Chemistry Equation Sheet

**Mary Millhollon, Richard H. Langley**



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In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Chemistry Equation Sheet**," compiled by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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Under Food Issue 8 Oct 17, 2019 — The mass balance is the quantity of incoming raw material against the quantity used in the resulting finished products, taking process waste and ... The Mass Balance Approach in Feedstock Substitution An established method to foster sustainability in existing infrastructure · Benefits of the Mass Balance Approach · Biomass balance and ChemCycling · ChemCycling ... 8. Mass Balance Mass-balance analysis may also be referred to as. “Material Flow Analysis” or “Substance Flow Analysis.” Table 8.1 provides several examples of possible inputs,. Mass Balance Approach in the Chemical Industry The mass balance Approach (MBA) is a process for determining the use of chemically recycled or bio-based feedstock in a final product when both recycled and ... BRC 3.9.2 Trace Exercise Sample Procedure to conduct a mass balance check · 1. Select a raw material lot number used in a finished product made within the last 6 months. · 2. Review storage ... UNDERSTANDING VULNERABILITY ASSESSMENT Table 6 provides examples of PRNs for different raw materials. Table 6 Priority ... Mass balance exercises at critical points in the supply chain – the mass ... ISSUE 8 FOOD SAFETY - Frequently Asked Questions - a worked example from the raw material supplier, which ... to conduct a mass balance test every 6 months for each claim or a single mass balance test every. 3 Pedrotti - Solution Manual for Introduction to Optics On Studocu you find all the lecture notes, summaries and study guides you need to pass your exams with better grades. Solution For Optics Pedrotti | PDF solution-for-optics-pedrotti[272] - Read book online for free. optics solution. Manual Introduction to Optics Pedrotti.pdf Manual Introduction to Optics Pedrotti.pdf. Manual Introduction to Optics ... Hecht Optics Solution Manual. 37 1 10MB Read ... Introduction To Optics 3rd Edition Textbook Solutions Access Introduction to Optics 3rd Edition solutions now. Our solutions are written by Chegg experts so you can be assured of the highest quality! Solution For Optics Pedrotti The microscope first focuses on the scratch using direct rays. Then it focuses on the image I2 formed in a two step process: (1) reflection from the bottom ... Introduction to Optics - 3rd Edition - Solutions and Answers Our resource for Introduction to Optics includes answers to chapter exercises, as well as detailed information to walk you through the process step by step. Introduction to Optics: Solutions Manual Title, Introduction to Optics: Solutions Manual. Authors, Frank L. Pedrotti, Leno S. Pedrotti. Edition, 2. Publisher, Prentice Hall, 1993. Optics Pedrotti Solution Manual Pdf Optics Pedrotti Solution Manual Pdf. INTRODUCTION Optics Pedrotti Solution Manual Pdf Copy. Manual Introduction To Optics Pedrotti PDF Manual Introduction to Optics Pedrotti.pdf - Free ebook download as PDF File (.pdf), Text File (.txt) or read book online for free. Solutions Manual for Introduction to Optics 3rd Edition ... Mar 25, 2022 - Solutions Manual for Introduction to Optics 3rd Edition by Pedrotti Check more at ...