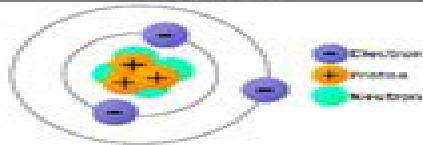
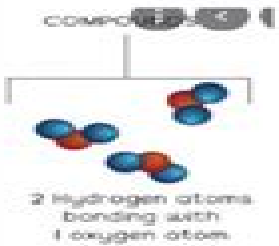


Recap Chemistry Questions

1. In the table below these are some terms you have heard of before. Use a diagram and example(s) to form a description of the following terms.

Terms	Diagram	Example(s)	Description
<i>Atom</i>		Hydrogen atom, helium atom, iron atom	An atom is made up of a nucleus consisting of neutrons and protons, and electrons moving quickly around the nucleus. It is the smallest part of a chemical element that can exist. Atoms are the building blocks of all matter. An atom is a particle of matter that uniquely defines a chemical element.
<i>Element</i>		Hydrogen, helium, iron, gold, silver	A pure substance consisting only of atoms that all have the same numbers of protons in their nuclei. Elements are shown on the periodic table. Combinations of elements can create compounds. An element is determined by the number of protons it has in its nucleus. A pure substance which cannot be broken down or changed into another substance using chemical means, contains only one type of atom.
<i>Compound</i>		Water (H ₂ O), methane, table salt	A substance composed of identical molecules consisting of atoms of two or more chemical elements. A pure substance formed from two or more elements chemically joined in fixed proportions.
<i>Mixture</i>		Orange juice, salt water.	The physical combination of two or more substances in which the identities are retained and are mixed in the form of solutions, suspensions and colloids. An impure substance made up of two or more different substances which

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