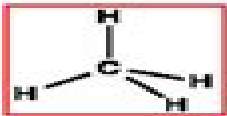
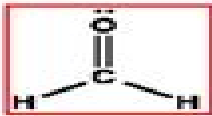



## Hydrocarbons Guided Learning Solutions

### Part 1: Carbon Bonds

- Is carbon a metal or a non-metal? non-metal
- Write the electron configuration of carbon.  $1s^2 2s^2 2p^2$
- Complete the following table:

	$\text{CH}_4$	$\text{H}_2\text{CO}$	$\text{HCN}$
Draw the Lewis structure (carbon is the central atom)			
Describe the molecular shape	<span style="color: red;">tetrahedral</span>	<span style="color: red;">trigonal planar</span>	<span style="color: red;">linear</span>
State the approximate bond angle	<span style="color: red;"><math>109.5^\circ</math></span>	<span style="color: red;"><math>120^\circ</math></span>	<span style="color: red;"><math>180^\circ</math></span>

- How many bonds do hydrogen and carbon form? C: 4    H: 1
- Define hydrocarbon.  
A compound containing carbon and hydrogen
- What do the terms saturated and unsaturated mean with respect to hydrocarbons?  
Saturated: The hydrocarbon has only single bonds  
Unsaturated: The hydrocarbon has one or more multiple bonds.

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**Robert R. Redfield**



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