

# CBSE PAPER-2024 (SET-1)

## CHEMISTRY

1. The molar ionic conductivities of  $\text{Ca}^{2+}$  and  $\text{Cl}^-$  are  $119.0$  and  $76.3 \text{ S cm}^2 \text{ mol}^{-1}$  respectively. The value of limiting molar conductivity of  $\text{CaCl}_2$  will be : [1]  
 (a)  $195.3 \text{ S cm}^2 \text{ mol}^{-1}$  (b)  $43.3 \text{ S cm}^2 \text{ mol}^{-1}$   
 (c)  $314.3 \text{ S cm}^2 \text{ mol}^{-1}$  (d)  $271.6 \text{ S cm}^2 \text{ mol}^{-1}$
2. Consider the following reaction : [1]



Identify A and B from the given options :

- (a) A – Methanol, B – Potassium formate (b) A – Ethanol, B – Potassium formate  
 (c) A – Methanal, B – Ethanol (d) A – Methanol, B – Potassium acetate
3. Which of the following acids represents Vitamin C ? [1]  
 (a) Saccharic acid (b) Gluconic acid  
 (c) Ascorbic acid (d) Benzoic acid
4. Rosenmund reduction is used for the preparation of Aldehydes. The catalyst used in this reaction is : [1]  
 (a)  $\text{Pd} - \text{BaSO}_4$  (b) Anhydrous  $\text{AlCl}_3$   
 (c) Iron (III) oxide (d)  $\text{HgSO}_4$
5. Which alkyl halide from the given options will undergo  $\text{S}_{\text{N}}1$  reaction faster ? [1]  
 (a)  $(\text{CH}_3)_3\text{C}-\text{Br}$  (b)  $(\text{CH}_3)_2\text{CH}-\text{Br}$  (c)  $\text{CH}_3-\text{CH}_2-\text{Br}$  (d)  $(\text{CH}_3)_3\text{C}-\text{CH}_2-\text{Br}$
6. From the elements of 3d series given below, which element shows the maximum number of oxidation states ? [1]  
 (a) Scandium (b) Manganese (c) Chromium (d) Titanium

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**Tao Wei**



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