#### WASSCE / WAEC MAY / JUNE 2006 CHEMISTRY THEORY PAPER

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|           | . 12   |                   |                 |
|-----------|--|-------------------|-----------------|
|           | PART B   |                   | and the same    |
|           | Essay<br>[ 100 marks ]   | 11.3              | 2 hours         |
|           | Answer four questions in all: three questions from Section I and   | LH <sub>C</sub> D |                 |
|           | one question from either Section II or Section III.  |                   |                 |
|           | All questions carry amai marky   |                   | - 3             |
|           | Credit will be given for clarity of expression and orderly presentation of me  | tiertal.          |                 |
|           | Section 1 Section 1  |                   | Service Control |
|           | FOR ALL CANDIDADES   |                   |                 |
|           | Answer three questions from this section.  |                   |                 |
| (4)       |  | 34.5              |                 |
|           | (i) Nuclear fission;   |                   |                 |
| 70.       | (ii) Nuclear fusion.   | 14:               | marks 1         |
|           | Alpha particle emission by <sup>233</sup> <sub>92</sub> U produces an element A. Beta particle e particle A produces another element B. Element B also undergoes alpha pe to produce <sup>227</sup> <sub>89</sub> Ac. Write balanced equations to represent the above statement  | market a first on | mission         |
| Carrier . | All the second s | 14                | murks ]         |
| cer       | The models below represent the filling of orbitals in an atom.   | arin may          | 171             |
|           | 1 11 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1   |                   | A STATE OF      |
|           | State which rule(s) is lace violated or obeyed by each model.  | (6)               | macka ]         |
| 649       | Explain why the boiling point of H <sub>2</sub> S with relative molecular mass of 34 is 1 of H <sub>2</sub> O with relative molecular mass of 18.  | ower ti           | on that         |
|           | Chip - report  | DAI               | mirrior 1 1     |
| (m).      | HCl is passed into each of the following solvents:   | 经被销               |                 |
|           | (i) water; -1 14 (1)   |                   |                 |
|           | (ii) methylbenzene.  |                   | 9               |
|           | I. State the effect of each solution on blue litmus paper. He warn to re-  | 174               | 7)              |
|           | II. Compare the electrical conductivities of the two solutions.  | The second second | narks I         |
| 20        | Zinc dust is added to copper (II) tetraexosulphate (VI) solution.  |                   |                 |
|           | State (i) what is observed:  | LL                | × 90            |
| 45        | (ii) the type of reaction that occurs.   | 1000              | nirki 1         |
| 154       | AD LODGING AT THE NEXT PAGE HEFO   | TACK.             | make 1          |
|           | VOIL ARE TOLD TO DO SO.  |                   |                 |

# **Chemistry Paper Answers Waec May June**

**Vera Liondas** 

#### **Chemistry Paper Answers Waec May June:**

School Science Practical Work in Africa Umesh Ramnarain,2020-06-01 School Science Practical Work in Africa presents the scope of research and practice of science practical work in African schools It brings together prominent science educators and researchers from Africa to share their experience and findings on pedagogical innovations and research informed practices on school science practical work The book highlights trends and patterns in the enactment and role of practical work across African countries Practical work is regarded as intrinsic to science teaching and learning and the form of practical work that is strongly advocated is inquiry based learning which signals a definite paradigm shift from the traditional teacher dominated to a learner centered approach The book provides empirical research on approaches to practical work contextual factors in the enactment of practical work and professional development in teaching practical work This book will be of great interest to academics researchers and post graduate students in the fields of science education and educational policy

Jos Educational Forum ,2009

Nigerian Journal of Science and Educational Research ,2007

Annual Conference Proceedings Science Teachers Association of Nigeria. Conference, 2006 **West African Journal of** Daily Graphic Ransford Tetteh, 2011-02-23 Report University of Reading. National Institute for **Education** ,1964 Research in Dairving, Shinfield, 1939 **Report** National Institute for Research in Dairving (Great Report ,1941 Britain).1941 West Africa ,1981-09 African Concord .1985 Report - National Institute for Research in Dairying National Institute for Research in Dairying (Great Britain),1941 Thisweek ,1987 **Education Sector Reform in** Nigeria David Nwoye Ezeh, Nkadi Onyegegbu, 2008 Happyhome ,1977 Drum ,1977 **Chemistry** Phillip John Buddle, J. S. Mackay, 1970 **Chemistry Questions and Answers** Kingsley Augustine, 2017-08-02 This is a chemistry book which is suitable for students in high schools or secondary schools It will also serve as a useful tool for students who are preparing for entrance examinations into colleges and universities Firs year students in the higher institutions taking courses in chemistry in will also find this book very useful Numerous topics in chemistry have been covered by the questions contained in this handy book At the end of the question section comes the answer section which supplies all answers to the questions Therefore students are thus presented with an effective means of self assessment whereby they can determine their individual strengths and revision needs A constructive review of this chemistry handbook on objective questions will be highly appreciated from buyers so as to give ideas to others who intend to purchase a copy of this book and also to be a form of advice for the author when revising the book **Chemistry** Vera Liondas, 1978 College Chemistry MCOs Arshad Igbal, 2017-08-29 College chemistry multiple choice questions has 1410 MCQs College chemistry guiz questions and answers MCQs on organic chemistry basic chemistry atomic structure chemical formulas chemical equations gas laws Charles s law Boyle s law inorganic chemistry MCQs with answers chemical science chemical reactions chemical bonding liquids and solids MCQs and quiz study guides for SAT ACT GAT GRE CLEP GED practice tests College chemistry multiple choice quiz

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