



Chemical Calculations

Multiple Choice Questions

- Q1. What is the M_r of copper (II) oxide (CuO)?
- 79.5
 - 37.0
 - 63.5
 - 39.75
- Q2. How many moles are there in 39.75g of CuO?
- 1 mole
 - 0.25 moles
 - 2 moles
 - 0.5 moles
- Q3. When copper (II) oxide and hydrochloric acid react together they produce copper (II) chloride and water. Which equation correctly represents this reaction?
- $\text{CuO} + \text{HCl} \rightarrow \text{CuCl} + \text{H}_2\text{O}$
 - $\text{Cu} + 2\text{HCl} \rightarrow \text{CuCl}_2 + \text{H}_2$
 - $\text{CuO} + \text{HCl} \rightarrow \text{CuCl}_2 + \text{H}_2\text{O}$
 - $\text{CuO} + 2\text{HCl} \rightarrow \text{CuCl}_2 + \text{H}_2\text{O}$
- Q4. What is the correct symbol equation for chromium (III) oxide reacting with aluminium?
- $\text{Cr}_2\text{O}_3 + \text{Al} \rightarrow \text{Cr} + \text{Al}_2\text{O}_3$
 - $\text{CrO} + \text{Al} \rightarrow \text{Cr} + \text{AlO}$
 - $\text{Cr}_2\text{O}_3 + 2 \text{Al} \rightarrow 2 \text{Cr} + \text{Al}_2\text{O}_3$
 - $\text{Cr}_2\text{O}_3 + \text{Al} \rightarrow 2 \text{Cr} + \text{AlO}_3$
- Q5. If 10.8g of aluminium metal is reacting in Q4 how many moles of aluminium metal are involved in the reaction?
- 2 moles
 - 0.4 moles
 - 1 mole
 - 0.2 moles
- Q6. In theory, how much mass of the chromium product could be produced in Q5?
- 30.4g
 - 1g
 - 0.4g
 - 20.8g
- Q7. What is the percentage yield of sodium oxide in the following reaction if 20g of sodium metal reacted and actually produced 14g of sodium oxide (Na₂O)? NB in theory this chemical reaction should have produced 27g.
- $$4 \text{Na} + \text{O}_2 \rightarrow 2 \text{Na}_2\text{O}$$
- 51.9%
 - 53.9%
 - 74.1%
 - 70.0%

Chemistry Sample Questions Multiple Choice 2013

Arshad Iqbal



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Hodges, 2023-07-03 This book is written for all science or engineering faculty who have ever found themselves baffled and frustrated by their undergraduate students lack of engagement and learning The author an experienced scientist faculty member and educational consultant addresses these issues with the knowledge of faculty interests constraints and day to day concerns in mind Drawing from the research on learning she offers faculty new ways to think about the struggles their science students face She then provides a range of evidence based teaching strategies that can make the time faculty spend in the classroom more productive and satisfying Linda Hodges reviews the various learning problems endemic to teaching science explains why they are so common and persistent and presents a digest of key ideas and strategies to address them based on the research she has undertaken into the literature on the cognitive sciences and education Recognizing that faculty have different views about teaching different comfort levels with alternative teaching approaches and are often

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Parliamentary Debates, House of the People India. Parliament. Lok Sabha,2013-03

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