

## Chemistry Worksheets Class 11 on Chapter 5 States of Matter- Set 1

**Q-1:** At 600 K, the vapour pressure of water in a closed container is 0.5 atm. If the container's volume is doubled, its vapour pressure at 600 K will be

- a) 0.2 atm
- b) 0.8 atm
- c) 0.6 atm
- d) 0.5 atm

**Q-2:** In the equation  $PV = RT$ , the value of R will depend on

- a) The nature of the gas
- b) The pressure of the gas
- c) The temperature of the gas
- d) Units of measurement

**Q-3:** Air at sea level is dense. This is the practical application of

- (a) Charles's law
- (b) Boyle's law
- (c) Dalton's law
- (d) Avogadro's law

**Q-4:** Triple point of water is

- a) 373.16 K
- b) 273.16 K
- c) 193.14 K
- d) 203.14 K

**Q-5:** The slope of the plot between  $PV$  and  $P$  at constant temperature is

- a) 1
- b) Zero
- c)  $\frac{1}{2}$
- d) None of the above

**Q-6:** While pumping air into a cycle tyre, a boy noticed that volume, as well as pressure, was increasing. Does Boyle's law fail here?

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