

# ATOMIC STRUCTURE

## RUTHERFORD'S MODEL

- Discovered that an atom is mostly made up of empty space
- Center of the atom is a small dense positively charged nucleus
- This nucleus holds the negatively charged particles in the atom

## BOHR'S MODEL

- There are 2 main pieces of an atom:
- Nucleus
- Electrons

## THE OLD MODEL OF THE ATOM



## WAVE-MECHANICAL MODEL

- Electrons are moving so fast that they look like a cloud. Orbitals are regions of most likely electron location



## WHAT IS TRUE TODAY...

- The nucleus is composed of 2 things:
  - PROTONS - positively charged particles
  - NEUTRONS - neutral charged particles
- Electrons are in a cloud - they are in orbitals
  - Electrons DO move around the nucleus in different patterns



Subatomic Particle	Location in the Atom	Charge	Weight
proton	center (in the nucleus)	positive	1 amu
neutron	center (in the nucleus)	neutral (no charge)	1 amu
electron	around the nucleus in clouds or orbitals	negative	Almost equal to neutrons (1/2)

### IONS

- If an atom gains or loses an electron, it is called an ion
- This gives the atom a charge:
  - Lose an electron, the atom becomes positive
  - Gain an electron, the atom becomes negative

### ISOTOPES

- An isotope is when an atom is physically the same, but different
- An isotope is still the same element, it is just a little different from every other atom of the same element
  - slightly different mass

# Chemistry Atomic Structure Guide

**Didier Musso**



## **Chemistry Atomic Structure Guide:**

**A Guide to the Literature of Chemistry** Evan Jay Crane, Austin McDowell Patterson, 1927      **Research in Education**, 1974      **Subject Guide to Books in Print**, 2001      *Readers' Guide*, 1912      Molecular Quantum Mechanics Peter W. Atkins, Ronald S. Friedman, 2011 This text unravels those fundamental physical principles which explain how all matter behaves It takes us from the foundations of quantum mechanics through quantum models of atomic molecular and electronic structure and on to discussions of spectroscopy and the electronic and magnetic properties of molecules      *Resources in Education*, 1984-10      **List of Books for School Libraries of the State of Oregon**, 1917      **Addison-Wesley Chemistry**, 1990      **The Guide to Nature**, 1922      **Educational Screen & Audio-visual Guide**, 1966      *Science Abstracts*, 1908      **Industrial Canada**, 1925      Study Guide [to] Chemistry, a Systematic Approach [by] Sisler, Dresdner, and Mooney William H. Myers, William T. Mooney, 1980      The English Catalogue of Books Sampson Low, 1926 Volumes for 1898 1968 include a directory of publishers      **Chemical Abstracts**, 1921      **Nature** Sir Norman Lockyer, 1909      Chemist and Druggist, 1927      General Chemistry James Kendall, Alexander Smith, 1927      *An Introduction to General Chemistry* William Martin Blanchard, 1928      *The Open Shelf*, 1919

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