

# AQA Chemistry

## GCSE Exam-style mark scheme

### C1 Atomic structure

GCSE Combined Science: Trilogy

11/10/19

Question	Answers	Extra Information	Mark	AO / Spec ref.
01.1	$2\text{Na} + \text{Cl}_2 \rightarrow 2\text{NaCl}$	Allow correct multiples; allow 1 mark for Na and NaCl	2	AO2 C1.1.1
01.2	<p><b>Level 3 (5–6 marks):</b> A full, clear description of the process is provided. The major steps are outlined in a logical method that would result in the required products being obtained.</p> <p><b>Level 2 (3–4 marks):</b> The substantive content of the process is present, but some steps are missing. The explanation may not be in a completely logical sequence.</p> <p><b>Level 1 (1–2 marks):</b> Simple statements relating to the process are made but with no logical structure.</p> <p><b>Level 0 (0 marks):</b> No relevant content.</p> <p><b>Indicative Content</b></p> <ul style="list-style-type: none"> <li>• add water</li> <li>• stir/warm mixture</li> <li>• filter to remove sand</li> <li>• sand washed to remove salt solution</li> <li>• sand dried in warm oven</li> <li>• filtrate/salt solution heated</li> <li>• suitable method for crystallisation, e.g., salt solution heated in evaporating basin on a water bath until crystals first appear on side of dish or half volume of solution evaporated</li> <li>• leave to crystallise in warm place or leave for a few days.</li> </ul> <p>This indicative content is not exhaustive, other creditworthy responses should be awarded marks as appropriate.</p>		6	AO1 C1.1.2
02	4 protons; 5 neutrons; 4 electrons; protons and neutrons in nucleus; electrons in energy levels/shells.	Allow electronic configuration is 2,2.	1 1 1 1 1	AO1×3 AO2×2 C1.1.4 C1.1.7
03.1	$\text{CO}_2$		1	AO3 C1.1.1

# Chemistry C1 Edexcel Mark Scheme May 2012

**Lingsheng Yao**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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