

Name	Alia Adennoun	Lab Day	Monday	Bench #	H01
Experiment #	3	Title	Periodic table and trends of elements		

### Prelab assignment

Sentence answers are not required for this assignment. Type formulas using correct subscripts. Formulas not typed with number of atoms as subscripts will not get full credit.

**Q1: Determine the formula for anhydrides of the following acids and bases:**

a)  $\text{Ba(OH)}_2$       b)  $\text{H}_2\text{SO}_3$       c)  $\text{HClO}_3$       d)  $\text{Fe(OH)}_3$

Acids and Bases	Anhydrides
$\text{Ba(OH)}_2$	$\text{BaO}$
$\text{H}_2\text{SO}_3$	$\text{SO}_2$
$\text{HClO}_3$	$\text{Cl}_2\text{O}_6$
$\text{Fe(OH)}_3$	$\text{Fe}_2\text{O}_3$

**Q2: Determine the formula of the parent acid or base parents of the following anhydrides:**

a)  $\text{N}_2\text{O}_5$       b)  $\text{SO}_3$       c)  $\text{MgO}$       d)  $\text{P}_2\text{O}_5$

Anhydrides	Parent acid or base
$\text{N}_2\text{O}_5$	$\text{HNO}_3$
$\text{SO}_3$	$\text{H}_2\text{SO}_4$
$\text{MgO}$	$\text{Mg(OH)}_2$
$\text{P}_2\text{O}_5$	$\text{H}_3\text{PO}_4$

**Q3: Write the names of the following compounds:**

a)  $\text{SO}_3$       b)  $\text{H}_2\text{SO}_3$       c)  $\text{HCl}$       d)  $\text{Ca(OH)}_2$       e)  $\text{HClO}_4$   
 f)  $\text{N}_2\text{O}_5$       g)  $\text{P}_4\text{O}_{10}$       h)  $\text{CO}_2$

Formula	Name
$\text{SO}_3$	sulfur trioxide
$\text{H}_2\text{SO}_3$	sulfurous acid
$\text{HCl}$	hydrochloric acid
$\text{Ca(OH)}_2$	calcium hydroxide
$\text{HClO}_4$	Perchloric acid
$\text{N}_2\text{O}_5$	dinitrogen pentaoxide
$\text{P}_4\text{O}_{10}$	tetraphosphorus pentaoxide
$\text{CO}_2$	carbon dioxide

# Chemistry Ch 19 Assessment Answers

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## **Table of Contents Chemistry Ch 19 Assessment Answers**

1. Understanding the eBook Chemistry Ch 19 Assessment Answers
  - The Rise of Digital Reading Chemistry Ch 19 Assessment Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Ch 19 Assessment Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an eBook Chemistry Ch 19 Assessment Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Ch 19 Assessment Answers
  - Personalized Recommendations
  - Chemistry Ch 19 Assessment Answers User Reviews and Ratings
  - Chemistry Ch 19 Assessment Answers and Bestseller Lists
5. Accessing Chemistry Ch 19 Assessment Answers Free and Paid eBooks
  - Chemistry Ch 19 Assessment Answers Public Domain eBooks
  - Chemistry Ch 19 Assessment Answers eBook Subscription Services
  - Chemistry Ch 19 Assessment Answers Budget-Friendly Options
6. Navigating Chemistry Ch 19 Assessment Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Chemistry Ch 19 Assessment Answers Compatibility with Devices
  - Chemistry Ch 19 Assessment Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Chemistry Ch 19 Assessment Answers
  - Highlighting and Note-Taking Chemistry Ch 19 Assessment Answers
  - Interactive Elements Chemistry Ch 19 Assessment Answers



8. Staying Engaged with Chemistry Ch 19 Assessment Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Chemistry Ch 19 Assessment Answers
9. Balancing eBooks and Physical Books Chemistry Ch 19 Assessment Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Chemistry Ch 19 Assessment Answers
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Chemistry Ch 19 Assessment Answers
  - Setting Reading Goals Chemistry Ch 19 Assessment Answers
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemistry Ch 19 Assessment Answers
  - Fact-Checking eBook Content of Chemistry Ch 19 Assessment Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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