

SOLUTIONS

9. A 1% solution of solute 'X' is isotonic with a 6% solution of sucrose (molar mass = 342 g mol^{-1}). The molar mass of solute 'X' is :
 (a) 34.2 g mol^{-1} (b) 57 g mol^{-1}
 (c) 114 g mol^{-1} (d) 3.42 g mol^{-1} **(1)**
8. On mixing 10 mL of acetone with 50 mL of chloroform, the total volume of the solution will be :
 (a) $< 60 \text{ mL}$ (b) $> 60 \text{ mL}$
 (c) $= 60 \text{ mL}$ (d) $= 100 \text{ mL}$ **(1)**
4. The system that forms maximum boiling azeotrope is :
 (a) CS_2 and acetone (b) Benzene and toluene
 (c) Ethanol and acetone (d) Chloroform and acetone **(1)**
5. Phenol dimerises in benzene having van't Hoff factor 0.54. Its degree of association is:
 (a) 0.54 (b) 0.46
 (c) 0.92 (d) 0.27
1. Low concentration of oxygen in the blood and tissues of people living at high altitudes is due to :
 (a) Low temperature (b) Low atmospheric pressure
 (c) High atmospheric pressure (d) High temperature **(1)**
4. For an electrolyte undergoing association in a solvent, the van't Hoff factor :
 (a) is always greater than one
 (b) has negative value
 (c) has zero value
 (d) is always less than one **(1)**
8. An azeotropic solution of two liquids has boiling point lower than either of them when it :
 (a) is saturated
 (b) shows positive deviation from Raoult's law
 (c) shows negative deviation from Raoult's law
 (d) shows no deviation from Raoult's law **(1)**
9. The relative lowering of vapour pressure of an aqueous solution containing non-volatile solute is 0.0225. The mole fraction of the non-volatile solute is :
 (a) 0.80 (b) 0.725
 (c) 0.15 (d) 0.0225 **(1)**
- 10*. A 1% solution of solute 'X' is isotonic with a 6% solution of sucrose (molar mass = 342 g mol^{-1}). The molar mass of solute 'X' is :
 (a) 34.2 g mol^{-1} (b) 57 g mol^{-1}
 (c) 114 g mol^{-1} (d) 3.42 g mol^{-1} **(1)**

Question numbers 13 to 16 are assertion reason types.

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