

Section 6.1: Organizing the Elements

- Group 1 on the periodic table is called **Alkali Metals**.
- Group 2 on the periodic table is called **Alkaline Earth Metals**.
- Elements in groups 3-12 are called **Transition Metals**.
- Group 17 on the periodic table is called **Halogens**.
- Group 18 on the periodic table is called **Noble Gases**.
- Each period on the periodic table corresponds to a **principal energy level**.
- The 2 series that make up the inner transition metals are **Lanthanides** and **Actinides**.
- Complete the following chart:

Element	Symbol	Group #	Period #	Group Name	Metal, Metalloid, Nonmetal
Zirconium	Zr	4	5	Transition Metals	Metal
Radium	Ra	2	7	Alkaline Earth Metals	Metal
Bismuth	Bi	15	6	Nitrogen Group	Metal
Gallium	Ga	13	4	Boron Group	Metal

- What are triads? **Sets of 3 elements with similar properties**. Whose idea were they? **Dobereiner**.
- The modern periodic table is arranged by increasing **atomic number**. Whose idea was this? **Moseley**.
- Mendeleev** developed the first periodic table, which was arranged by increasing **atomic mass**.
- Determine which of the following pairs will have similar properties (Yes or no):
 - Ca and Be? **Yes – same group (Group 2 – Alkaline Earth Metals)**
 - Sr and Y? **No – not the same group**
 - C and Ge? **Yes – same group (Group 14 – Carbon Group)**

Section 6.3: Periodic Trends

- As you go across a period, the atomic radius (size) **decreases** because **the attraction between protons and electrons increases, which makes the atom smaller (opposite of tug of war)**.
- As you go down a group, the atomic radius (size) **increases** because **the number of principal energy levels increases**.
- Circle the atom with the largest atomic radius and put a square around the atom with the smallest radius:

a.	<div style="border: 1px solid black; border-radius: 50%; padding: 2px;">Li</div>	<div style="border: 1px solid black; padding: 2px;">Ne</div>	F	B
b.	<div style="border: 1px solid black; padding: 2px;">N</div>	<div style="border: 1px solid black; border-radius: 50%; padding: 2px;">Bi</div>	S	P
c.	Na	<div style="border: 1px solid black; border-radius: 50%; padding: 2px;">Sr</div>	Al	<div style="border: 1px solid black; padding: 2px;">Cl</div>
- Ionization energy is the energy required to **remove** an electron.

Chemistry Chapter 6 1 Organizing The Elements

Worksheet Answers

RS Peters



Chemistry Chapter 6 1 Organizing The Elements Worksheet Answers:

Concepts of Earth Science & Chemistry Parent Lesson Plan John Hudson Tiner, 2013-08-26 Concepts of Earth and Chemistry Course Description This is the suggested course sequence that allows one core area of science to be studied per semester You can change the sequence of the semesters per the needs or interests of your student materials for each semester are independent of one another to allow flexibility Semester 1 Earth Blending a creationism perspective of history with definitions of terms and identification of famous explorers scientists etc this book gives students an excellent initial knowledge of people and places encouraging them to continue their studies in depth Semester 2 Chemistry Chemistry is an amazing branch of science that affects us every day yet few people realize it or even give it much thought Without chemistry there would be nothing made of plastic there would be no rubber tires no tin cans no televisions no microwave ovens or something as simple as wax paper This book presents an exciting and intriguing tour through the realm of chemistry as each chapter unfolds with facts and stories about the discoveries of discoverers Find out why pure gold is not used for jewelry or coins Join Humphry Davy as he made many chemical discoveries and learn how they shortened his life See how people in the 1870s could jump over the top of the Washington Monument Exploring the World of Chemistry brings science to life and is a wonderful learning tool with many illustrations and biographical information **Survey of Science History & Concepts**

Parent Lesson Plan, 2013-08-01 Survey of Science History Concepts Course Description Students will study four areas of science Scientific Mathematics Physics Biology and Chemistry Students will gain an appreciation for how each subject has affected our lives and for the people God revealed wisdom to as they sought to understand Creation Each content area is thoroughly explored giving students a good foundation in each discipline Semester 1 Math and Physics Numbers surround us Just try to make it through a day without using any It s impossible telephone numbers calendars volume settings shoe sizes speed limits weights street numbers microwave timers TV channels and the list goes on and on The many advancements and branches of mathematics were developed through the centuries as people encountered problems and relied upon math to solve them It s amazing how ten simple digits can be used in an endless number of ways to benefit man The development of these ten digits and their many uses is the fascinating story in Exploring the World of Mathematics Physics is a branch of science that many people consider to be too complicated to understand John Hudson Tiner puts this myth to rest as he explains the fascinating world of physics in a way that students can comprehend Did you know that a feather and a lump of lead will fall at the same rate in a vacuum Learn about the history of physics from Aristotle to Galileo to Isaac Newton to the latest advances Discover how the laws of motion and gravity affect everything from the normal activities of everyday life to launching rockets into space Learn about the effects of inertia first hand during fun and informative experiments Exploring the World of Physics is a great tool for student who want to have a deeper understanding of the important and interesting ways that physics affects our lives Semester 2 Biology and Chemistry The field of biology focuses on living things from the

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Table of Contents Chemistry Chapter 6 1 Organizing The Elements Worksheet Answers

1. Understanding the eBook Chemistry Chapter 6 1 Organizing The Elements Worksheet Answers
 - The Rise of Digital Reading Chemistry Chapter 6 1 Organizing The Elements Worksheet Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Chapter 6 1 Organizing The Elements Worksheet Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an eBook Chemistry Chapter 6 1 Organizing The Elements Worksheet Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Chapter 6 1 Organizing The Elements Worksheet Answers
 - Personalized Recommendations
 - Chemistry Chapter 6 1 Organizing The Elements Worksheet Answers User Reviews and Ratings
 - Chemistry Chapter 6 1 Organizing The Elements Worksheet Answers and Bestseller Lists
5. Accessing Chemistry Chapter 6 1 Organizing The Elements Worksheet Answers Free and Paid eBooks
 - Chemistry Chapter 6 1 Organizing The Elements Worksheet Answers Public Domain eBooks
 - Chemistry Chapter 6 1 Organizing The Elements Worksheet Answers eBook Subscription Services
 - Chemistry Chapter 6 1 Organizing The Elements Worksheet Answers Budget-Friendly Options
6. Navigating Chemistry Chapter 6 1 Organizing The Elements Worksheet Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemistry Chapter 6 1 Organizing The Elements Worksheet Answers Compatibility with Devices
 - Chemistry Chapter 6 1 Organizing The Elements Worksheet Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry Chapter 6 1 Organizing The Elements Worksheet Answers
 - Highlighting and Note-Taking Chemistry Chapter 6 1 Organizing The Elements Worksheet Answers
 - Interactive Elements Chemistry Chapter 6 1 Organizing The Elements Worksheet Answers

8. Staying Engaged with Chemistry Chapter 6 1 Organizing The Elements Worksheet Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Chapter 6 1 Organizing The Elements Worksheet Answers
9. Balancing eBooks and Physical Books Chemistry Chapter 6 1 Organizing The Elements Worksheet Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Chapter 6 1 Organizing The Elements Worksheet Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chemistry Chapter 6 1 Organizing The Elements Worksheet Answers
 - Setting Reading Goals Chemistry Chapter 6 1 Organizing The Elements Worksheet Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemistry Chapter 6 1 Organizing The Elements Worksheet Answers
 - Fact-Checking eBook Content of Chemistry Chapter 6 1 Organizing The Elements Worksheet Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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