

# SAMPLE PAPER (2023 -24)

## CHEMISTRY THEORY (043)

Max. Marks:70

Time: 3 hours

### General Instructions:

Read the following instructions carefully.

- (a) There are **33** questions in this question paper with internal choice.
  - (b) SECTION A consists of 16 multiple -choice questions carrying 1 mark each.
  - (c) SECTION B consists of 5 short answer questions carrying 2 marks each.
  - (d) SECTION C consists of 7 short answer questions carrying 3 marks each.
  - (e) SECTION D consists of 2 case- based questions carrying 4 marks each.
  - (f) SECTION E consists of 3 long answer questions carrying 5 marks each.
  - (g) All questions are compulsory.
  - (h) Use of log tables and calculators is not allowed.
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### SECTION A

The following questions are multiple -choice questions with one correct answer. Each question carries 1 mark. There is no internal choice in this section.

- 1. Which of the following solutions will have the highest conductivity at 298 K?
  - (a) 0.01 M HCl solution
  - (b) 0.1 M HCl solution
  - (c) 0.01 M  $\text{CH}_3\text{COOH}$  solution
  - (d) 0.1 M  $\text{CH}_3\text{COOH}$  solution

- 2.  $\text{A} + \text{B} \xrightarrow{\text{dil NaOH}}$  

Identify A and B:

- (a) A = 1-phenylethanal, B = acetophenone
  - (b) A = Benzophenone, B = formaldehyde
  - (c) A = Benzaldehyde, B = Acetophenone
  - (d) A = Benzophenone, B = Acetophenone
- 3. The vitamins which can be stored in our body are:
  - (a) Vitamin A, B, D and E
  - (b) Vitamin A, C, D and K
  - (c) Vitamin A, B, C and D
  - (d) Vitamin A, D, E and K
- 4. What is IUPAC name of the ketone A, which undergoes iodo form reaction to give  $\text{CH}_3\text{CH}=\text{C}(\text{CH}_3)\text{COONa}$  and yellow precipitate of  $\text{CHI}_3$ ?
  - (a) 3-Methylpent-3-en-2-one
  - (b) 3-Methylbut-2-en- one
  - (c) 2, 3-Dimethylethanone
  - (d) 3-Methylpent-4-one
- 5. Which of the following is not correct?
  - (a) In haloarenes, the electron pairs on halogen atom are in conjugation with  $\pi$ -electrons of the ring.
  - (b) The carbon-magnesium bond is covalent and non-polar in nature.
  - (c) During  $\text{S}_{\text{N}}1$  reaction, the carbocation formed in the slow step being  $\text{sp}^2$  hybridised is planar.
  - (d) Out of  $\text{CH}_2=\text{CH}-\text{Cl}$  and  $\text{C}_6\text{H}_5\text{CH}_2\text{Cl}$ ,  $\text{C}_6\text{H}_5\text{CH}_2\text{Cl}$  is more reactive towards  $\text{S}_{\text{N}}1$  reaction

# Chemistry F322014 Question Paper

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