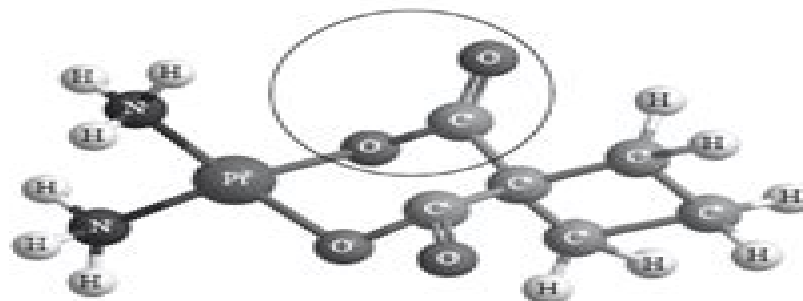


## HL Paper 2

Carboplatin used in the treatment of lung cancer has the following three-dimensional structure.



Elemental platinum has electrons occupying s, p, d and f atomic orbitals.

a. Identify the name of the functional group circled in the structure of carboplatin.

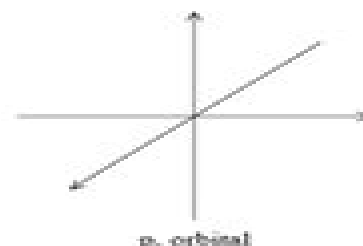
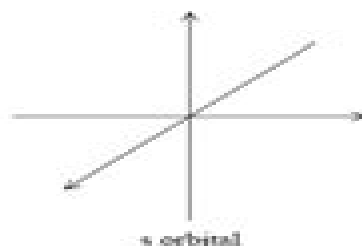
[1]

b. State the type of bonding between platinum and nitrogen in carboplatin.

[1]

c.i. Draw the shape of an s orbital and a  $p_x$  orbital. Label the x, y and z axes on each diagram.

[2]



c.ii. State the maximum number of orbitals in the  $n = 4$  energy level.

[1]

d. A number of ruthenium-based anti-cancer drugs have also been developed. State the full electron configuration of the ruthenium(II) ion,  $\text{Ru}^{2+}$ .

[1]

e. Iron is in the same group in the periodic table as ruthenium.

[1]

Construct the orbital diagram (using the arrow-in-box notation) for iron, showing the electrons in the  $n = 3$  and  $n = 4$  energy levels only and label each sub-level on the diagram.



3s



3p



3d



4s

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**Darrell D. Ebbing**



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