

# Introduction to Chemistry

## Section 1.1 The Stories of Two Chemicals

In your notebook, read aloud the entire paper.

Use each of the terms below just once to complete the passage.

stratosphere  
atmosphere

oxygen gas  
troposphere

ozone  
ultraviolet radiation

ozone hole

Earth's (1) \_\_\_\_\_ is made up of several layers. The air we breathe makes up the lowest level. This layer is called the (2) \_\_\_\_\_. The next layer up is called the (3) \_\_\_\_\_. This level contains a protective (4) \_\_\_\_\_ layer.

Ozone holes occur when (5) \_\_\_\_\_ is removed by ultraviolet radiation in the upper part of the atmosphere. The ozone forms a layer around Earth, which absorbs (6) \_\_\_\_\_. Without ozone, you are more likely to get a sunburn or possibly skin cancer. The thinning of the ozone layer, called the (7) \_\_\_\_\_, is a problem because without ozone all organisms on Earth are subject to more than one much radiation.

In your notebook, read aloud chlorofluorocarbons.

For each statement below, write true or false.

- \_\_\_\_\_ (A) CFC is another name for a chlorofluorocarbon.
- \_\_\_\_\_ (B) CFCs are made up of carbon, fluorine, and oxygen.
- \_\_\_\_\_ (C) All CFCs are synthetic chemicals.
- \_\_\_\_\_ (D) CFCs usually react easily with other chemicals.
- \_\_\_\_\_ (E) CFCs were developed as replacement for toxic refrigerants.

# Chemistry Matter And Change Chapter 1study Guide

**SB Merriam**



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## **Table of Contents Chemistry Matter And Change Chapter 1study Guide**

1. Understanding the eBook Chemistry Matter And Change Chapter 1study Guide
  - The Rise of Digital Reading Chemistry Matter And Change Chapter 1study Guide
  - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Matter And Change Chapter 1study Guide
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chemistry Matter And Change Chapter 1study Guide
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Matter And Change Chapter 1study Guide
  - Personalized Recommendations
  - Chemistry Matter And Change Chapter 1study Guide User Reviews and Ratings
  - Chemistry Matter And Change Chapter 1study Guide and Bestseller Lists

5. Accessing Chemistry Matter And Change Chapter 1study Guide Free and Paid eBooks
  - Chemistry Matter And Change Chapter 1study Guide Public Domain eBooks
  - Chemistry Matter And Change Chapter 1study Guide eBook Subscription Services
  - Chemistry Matter And Change Chapter 1study Guide Budget-Friendly Options
6. Navigating Chemistry Matter And Change Chapter 1study Guide eBook Formats
  - ePub, PDF, MOBI, and More
  - Chemistry Matter And Change Chapter 1study Guide Compatibility with Devices
  - Chemistry Matter And Change Chapter 1study Guide Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Chemistry Matter And Change Chapter 1study Guide
  - Highlighting and Note-Taking Chemistry Matter And Change Chapter 1study Guide
  - Interactive Elements Chemistry Matter And Change Chapter 1study Guide
8. Staying Engaged with Chemistry Matter And Change Chapter 1study Guide
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Chemistry Matter And Change Chapter 1study Guide
9. Balancing eBooks and Physical Books Chemistry Matter And Change Chapter 1study Guide
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Chemistry Matter And Change Chapter 1study Guide
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Chemistry Matter And Change Chapter 1study Guide
  - Setting Reading Goals Chemistry Matter And Change Chapter 1study Guide
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemistry Matter And Change Chapter 1study Guide
  - Fact-Checking eBook Content of Chemistry Matter And Change Chapter 1study Guide
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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