

CHAPTER 3 CHAPTER ASSESSMENT**MATTER AND CHANGE****Reviewing Vocabulary**

Match the definition in Column A with the term in Column B.

Column A

- _____ 1. The outermost electrons of an atom
- _____ 2. The center of an atom
- _____ 3. Hot, highly ionized, electrically conducting gas
- _____ 4. The smallest particle of an element that retains that element's characteristics
- _____ 5. The change from a gas to a liquid
- _____ 6. A homogeneous mixture
- _____ 7. The change of one or more substances into other substances
- _____ 8. The attractive force between two ions of opposite charge
- _____ 9. A tiny particle with a negative electrical charge
- _____ 10. The average of the mass numbers of an element's isotopes
- _____ 11. The spontaneous process through which unstable nuclei emit radiation

Column B

- a. atom
- b. atomic mass
- c. chemical reaction
- d. condensation
- e. electron
- f. ionic bond
- g. solution
- h. nucleus
- i. plasma
- j. valence electrons
- k. radioactive decay

In the space at the left, write the word or phrase in parentheses that makes the statement correct.

- _____ 12. The combined number of protons and neutrons is the (atomic number, mass number).
- _____ 13. An (energy level, isotope) represents the area in an atom where an electron is most likely to be found.
- _____ 14. (An ionic bond, A covalent bond) is the attraction of two atoms for a shared pair of electrons that holds the atoms together.
- _____ 15. An atom that gains or loses an electron is a charged particle called (an ion, a molecule).
- _____ 16. (An element, A compound) is a substance that cannot be broken down into simpler substances.
- _____ 17. A tiny particle that has mass and a positive electrical charge is a (neutron, proton).
- _____ 18. (Acids, Bases) are solutions characterized by the formation of hydroxide ions (OH^-).
- _____ 19. The process of changing from a liquid into a gas is called (evaporation, sublimation).

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Table of Contents Chemistry Matter And Change Chapter 3 Answers

1. Understanding the eBook Chemistry Matter And Change Chapter 3 Answers
 - The Rise of Digital Reading Chemistry Matter And Change Chapter 3 Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Matter And Change Chapter 3 Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Matter And Change Chapter 3 Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Matter And Change Chapter 3 Answers
 - Personalized Recommendations

- Chemistry Matter And Change Chapter 3 Answers User Reviews and Ratings
- Chemistry Matter And Change Chapter 3 Answers and Bestseller Lists
- 5. Accessing Chemistry Matter And Change Chapter 3 Answers Free and Paid eBooks
 - Chemistry Matter And Change Chapter 3 Answers Public Domain eBooks
 - Chemistry Matter And Change Chapter 3 Answers eBook Subscription Services
 - Chemistry Matter And Change Chapter 3 Answers Budget-Friendly Options
- 6. Navigating Chemistry Matter And Change Chapter 3 Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemistry Matter And Change Chapter 3 Answers Compatibility with Devices
 - Chemistry Matter And Change Chapter 3 Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry Matter And Change Chapter 3 Answers
 - Highlighting and Note-Taking Chemistry Matter And Change Chapter 3 Answers
 - Interactive Elements Chemistry Matter And Change Chapter 3 Answers
- 8. Staying Engaged with Chemistry Matter And Change Chapter 3 Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Matter And Change Chapter 3 Answers
- 9. Balancing eBooks and Physical Books Chemistry Matter And Change Chapter 3 Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Matter And Change Chapter 3 Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chemistry Matter And Change Chapter 3 Answers
 - Setting Reading Goals Chemistry Matter And Change Chapter 3 Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemistry Matter And Change Chapter 3 Answers
 - Fact-Checking eBook Content of Chemistry Matter And Change Chapter 3 Answers

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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