

2.6 - Periodic Trends worksheet

Name _____

I. Short answer

1. Label each group on the periodic table.
2. Explain what valence electrons are.
3. Explain what makes an element stable?

II. Answer the questions using the symbols on the periodic table

1. What elements are in period 5?

What element(s) is a

2 halogens

3. Ionthamides

alkaline earth metals

Snoble gases

transition metals

Alkali metals

References

What elements are in the sublevel

98

107

Age Group	Percentage
18-24	~15%
25-34	~25%
35-44	~35%
45-54	~45%
55-64	~55%
65+	~65%

126

What elements are

13. metals.

14. nonmetals

15. metalloids,

16. Which element is the most reactive nonmetal?

17. Which element is the most reactive metal?

18. Which element below is the least reactive?

19. Which element has the largest radius in period 4?

20. Which element has the largest radius in the Alkali metals family?

21. Which element has the highest ionization energy in period four?

22. Which elements has the second largest atomic number?

23. Which element has the largest atomic mass?

24. Which element is located at period 4, group 15?

25. Which element is located at period 5, group 2?

27 Which elements will form +1 ions?

28. Which group of metals will have the same properties

29 Which group of nonmetals will have the same properties

30. On the back, define all chapter 5, section 3 vocabulary terms.

[illegible][illegible]

Chemistry Periodic Assessment 2 Answer Key

**Spencer L. Seager, Michael R.
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