

# Test Bank for Chemistry: Structure and Properties, 2nd Edition, Nivaldo J. Tro

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## Chemistry: Structure & Properties, 2e (Tro) Chapter 1: Atoms

### 1.1 Multiple Choice Questions

1) An element has three stable isotopes with masses of 27.977 amu, 28.976 amu, and 29.973 amu. The heaviest two isotopes have an abundance of 4.68% and 3.09%, respectively. What is the mass of the element?

- A) 27.684 amu
- B) 29.251 amu
- C) 28.985 amu
- D) 30.107 amu
- E) 28.991 amu

Answer: C

Diff: 3 Var: 1 Page Ref: 1.9

LO: 1.6

Global: G4

2) All of the following will lead to a homogeneous mixture EXCEPT

- A) sugar and coffee.
- B) mud and water.
- C) salt and water.
- D) ice tea and lemonade.
- E) vinegar and water.

Answer: B

Diff: 1 Var: 1 Page Ref: 1.2

LO: 1.1

Global: G2

3) How are gases different from solids and liquids?

- A) Gases can only be made up of atoms.
- B) The particles in a gas attract each other much more strongly than in solids and liquids.
- C) Gases are compressible.
- D) Only gases can take the shape of their container.
- E) Gases are colorless.

Answer: C

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**P. Austin Nuttall, James Henry Murray**



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